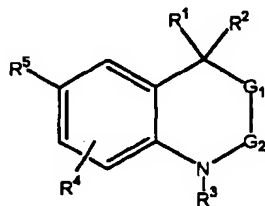




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(54) Title: QUINAZOLINONE AND BENZOXAZINE DERIVATIVES AS PROGESTERONE RECEPTOR MODULATORS



(I)

## (57) Abstract

This invention provides compounds which are agonists and antagonists of the progesterone receptor having general structure (I): wherein  $R^1$  and  $R^2$  are independently selected from H,  $COR^A$ , or  $NR^BCOR^A$ , or optionally substituted alkyl, alkenyl, alkynyl, cycloalkyl, aryl, or heterocyclic moieties; or  $R^1$  and  $R^2$  are fused to form: 3 to 8 membered spirocyclic alkyl, alkenyl or heterocyclic rings;  $R^A$  is H or optionally substituted alkyl, aryl, alkoxy, or aminoalkyl groups;  $R^B$  is H,  $C_1$  to  $C_3$  alkyl or substituted  $C_1$  to  $C_3$  alkyl;  $R^3$  is H, OH,  $NH_2$ ,  $COR^C$  or optionally substituted alkyl, alkenyl, or alkynyl;  $R^C$  is H or optionally substituted alkyl, aryl, alkoxy, or aminoalkyl;  $R^4$  is H, halogen, CN,  $NO_2$ , or optionally substituted alkyl, alkynyl, alkoxy, amino or aminoalkyl;  $R^5$  is an optionally substituted benzene or five or six membered ring with 1, 2, or 3 heteroatoms selected from O, S, SO,  $SO_2$  or  $NR^6$ ;  $R^6$  is H or  $C_1$  to  $C_3$  alkyl;  $G_1$  is O,  $NR_7$ , or  $CR_7R_8$ ;  $G_2$  is CO, CS, or  $CR_7R_8$ ; provided that when  $G_1$  is O,  $G_2$  is  $CR_7R_8$ , and  $G_1$  and  $G_2$  cannot both be  $CR_7R_8$ ;  $R_7$  and  $R_8$  are H or an optionally substituted alkyl, aryl, or heterocyclic moiety; or pharmaceutically acceptable salt thereof, and methods using these compounds in mammals as agonists or antagonists of the progesterone receptor.

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**QUINAZOLINONE AND BENZOXAZINE DERIVATIVES AS  
PROGESTERONE RECEPTOR MODULATORS**

5

**Field of the Invention**

This invention relates to compounds which are agonists and antagonists of the progesterone receptor, their preparation and utility.

**Background of the Invention**

10

Intracellular receptors (IR) form a class of structurally related gene regulators known as "ligand dependent transcription factors" (R. M. Evans, *Science*, **240**, 889, 1988). The steroid receptor family is a subset of the IR family, including progesterone receptor (PR), estrogen receptor (ER), androgen receptor (AR), glucocorticoid receptor (GR), and mineralocorticoid receptor (MR).

15

The natural hormone, or ligand, for the PR is the steroid progesterone, but synthetic compounds, such as medroxyprogesterone acetate or levonorgestrel, have been made which also serve as ligands. Once a ligand is present in the fluid surrounding a cell, it passes through the membrane *via* passive diffusion, and binds to the IR to create a receptor/ligand complex. This complex binds to specific gene promoters present in the cell's DNA. Once bound to the DNA the complex modulates the production of mRNA and protein encoded by that gene.

20

A compound that binds to an IR and mimics the action of the natural hormone is termed an agonist, whilst a compound which inhibits the effect of the hormone is an antagonist.

25

PR agonists (natural and synthetic) are known to play an important role in the health of women. PR agonists are used in birth control formulations, typically in the presence of an ER agonist. ER agonists are used to treat the symptoms of menopause, but have been associated with a proliferative effect on the uterus that can lead to an

- 2 -

increased risk of uterine cancers. Co-administration of a PR agonist reduces or ablates that risk.

PR antagonists may also be used in contraception. In this context they may be administered alone (Ulmann, et al, *Ann. N.Y. Acad. Sci.*, **261**, 248, 1995), in  
5 combination with a PR agonist (Kekkonen, et al, *Fertility and Sterility*, **60**, 610, 1993) or in combination with a partial ER antagonist such as tamoxifen (WO 96/19997 A1 July 4, 1996).

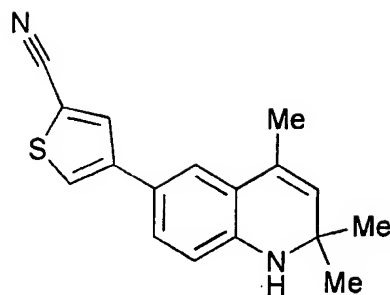
PR antagonists may also be useful for the treatment of hormone dependent breast cancers (Horwitz, et al, *Horm. Cancer*, 283, pub: Birkhaeuser, Boston, Mass.,  
10 ed. Vedeckis) as well as uterine and ovarian cancers. PR antagonists may also be useful for the treatment of non-malignant chronic conditions such as fibroids (Murphy, et al, *J. Clin. Endo. Metab.*, **76**, 513, 1993) and endometriosis (Kettel, et al, *Fertility and Sterility*, **56**, 402, 1991).

PR antagonists may also be useful in hormone replacement therapy for post  
15 menopausal patients in combination with a partial ER antagonist such as tamoxifen (US 5719136).

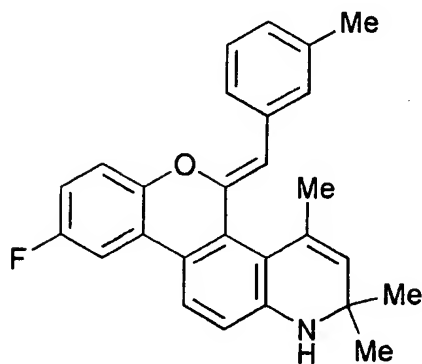
PR antagonists, such as mifepristone and onapristone, have been shown to be effective in a model of hormone dependent prostate cancer, which may indicate their utility in the treatment of this condition in men (Michna, et al, *Ann. N.Y. Acad. Sci.*,  
20 **761**, 224, 1995).

Jones, et al, (U.S. Patent No. 5,688,810) describe the PR antagonist dihydroquinoline **1**.

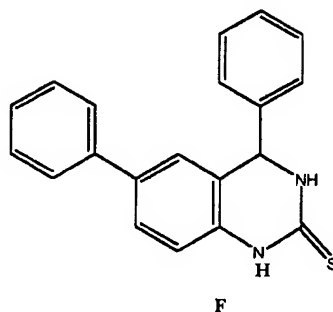
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**1**

Jones, *et al*, described the enol ether **2** (U.S. Patent No. 5,693,646) as a PR ligand.

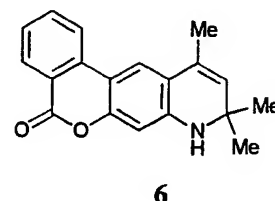
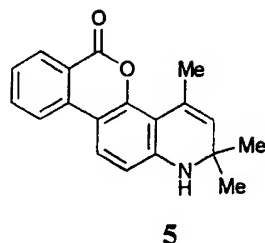
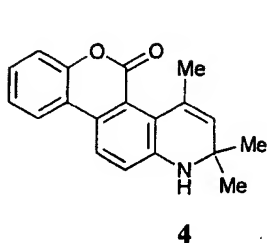
**2**

Jones, *et al*, described compound **3** (U.S. Patent No. 5,696,127) as a PR ligand.

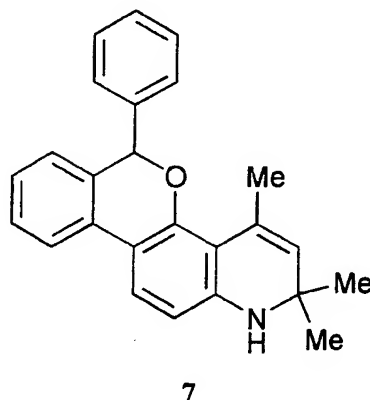
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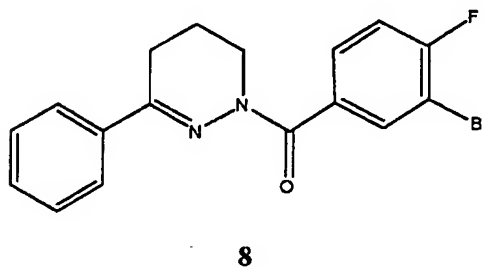
Zhi, *et al.*, described lactones 4, 5 and 6 as PR antagonists (*J. Med. Chem.*, 41, 291, 1998).



5 Zhi, *et al.*, described the ether 7 as a PR antagonist (*J. Med. Chem.*, 41, 291, 1998).

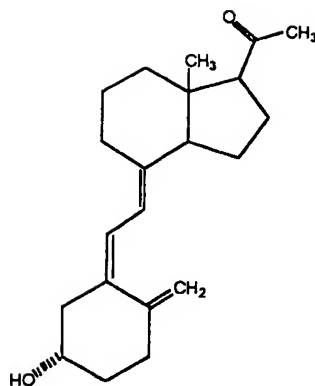


10 Combs, *et al.*, disclosed the amide 8 as a ligand for the PR (*J. Med. Chem.*, 38, 4880, 1995).



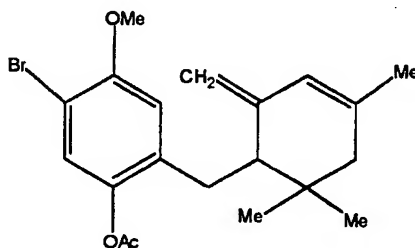
15 Perlman, *et al.*, described the vitamin D analog 9 as a PR ligand (*Tet. Letters*, 35, 2295, 1994).

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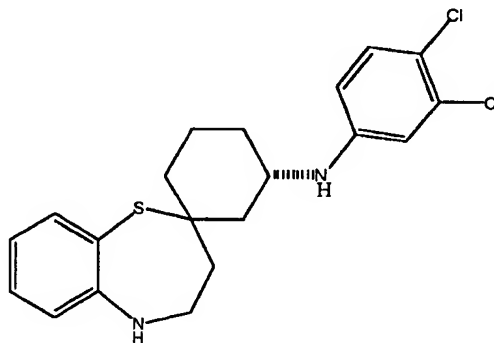
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Hamann, *et al*, described the PR antagonist **10** (*Ann. N.Y. Acad. Sci.*, **761**, 383, 1995).



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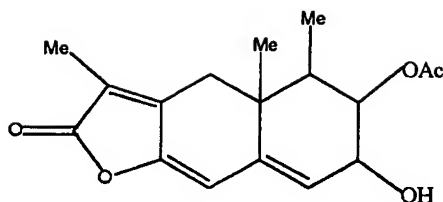
Chen, *et al*, described the PR antagonist **11** (Chen, *et al*, POI-37, 16<sup>th</sup> Int. Cong. Het. Chem., Montana, 1997).



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Kurihari, *et. al.*, described the PR ligand **12** (*J. Antibiotics*, **50**, 360, 1997).

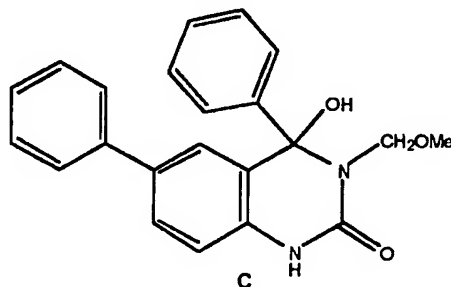
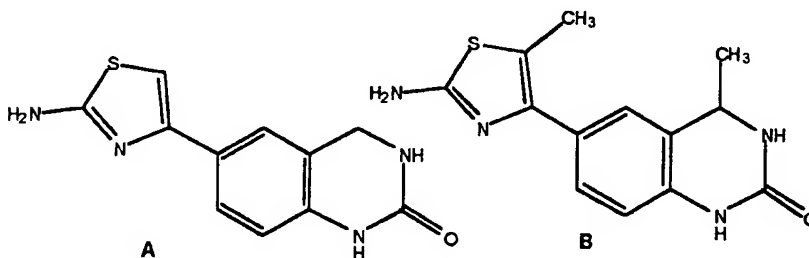
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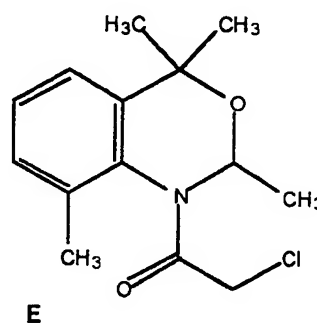
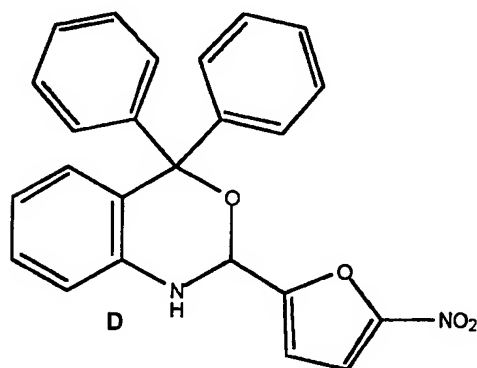
A number of publications reported the synthesis and utilities of benzodiazinones and benzoxazines. However, none of examples in this literature contained substituents necessary for the compounds to be active as progesterone receptor modulators. Included in this literature is the patent by Kubla et al. (US  
10 4666913) which claimed that the compound such as A and B could be used as cardiotonic agents. Ning et al. reported the synthesis of quinazolinones such as C.



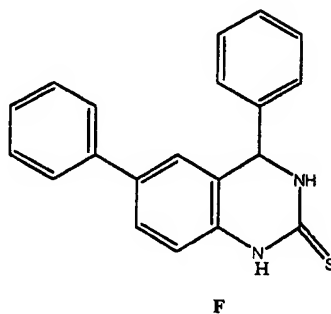


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Other prior art close to this invention is the literature which disclosed the benzoxazines. Among these publications, Gromachevskaya et al. (Chem. Heterocycl. Compd. (N.Y.), 33(10), 1209-1214 (1998)) studied the bromination process of certain benzoxazines such as compound D. Kobzina et al. (U.S. Patent No. 3,917,592) claimed that compounds such as E can be used as a herbicidal agent.



Pflegel et al. (Pharmazie, 37(10), 714-717(1982)) disclosed quinazolin-2-thiones such as compound F in their study of polarography of heterocyclics. No activity of the compound F was mentioned.



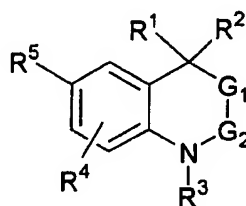
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**Description of the invention**

The compounds of this invention have been shown to act as competitive inhibitors of progesterone binding to the PR and act as agonists and/or antagonists in functional models, either/or *in-vitro* and *in-vivo*. These compounds may be used for  
 5 contraception, in the treatment of fibroids, endometriosis, breast, uterine, ovarian and prostate cancer, and post menopausal hormone replacement therapy.

The compounds in the present invention contain a pendent aromatic substituent. The aromatic substituents proved to be critical for the resultant compounds being active as progesterone receptor modulators and have broad  
 10 structural diversity which may consists of aryl, substituted aryl, heteroaryl or substituted heteroaryl group.

This invention provides compounds of Formula I having the structure:



I

15 wherein:

R<sup>1</sup>, R<sup>2</sup> are independent substituents selected from the group which includes H, C<sub>1</sub> to C<sub>6</sub> alkyl, substituted C<sub>1</sub> to C<sub>6</sub> alkyl, C<sub>2</sub> to C<sub>6</sub> alkenyl, substituted C<sub>2</sub> to C<sub>6</sub> alkenyl, C<sub>2</sub> to C<sub>6</sub> alkynyl, substituted C<sub>2</sub> to C<sub>6</sub> alkynyl, C<sub>3</sub> to C<sub>8</sub> cycloalkyl, substituted C<sub>3</sub> to C<sub>8</sub> cycloalkyl, aryl, substituted aryl, heterocyclic, substituted heterocyclic, COR<sup>A</sup>, or  
 20 NR<sup>B</sup>COR<sup>A</sup>;

or R<sup>1</sup> and R<sup>2</sup> are fused to form:

- a) an optionally substituted 3 to 8 membered spirocyclic alkyl ring;
- b) an optionally substituted 3 to 8 membered spirocyclic alkenyl; or

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c) an optionally substituted 3 to 8 membered heterocyclic ring containing one to three heteroatoms from the group including O, S and N; the spirocyclic rings of a), b) and c) being optionally substituted by from 1 to 4 groups selected from fluorine, C<sub>1</sub> to C<sub>6</sub> alkyl, C<sub>1</sub> to C<sub>6</sub> alkoxy, C<sub>1</sub> to C<sub>6</sub> thioalkyl, -CF<sub>3</sub>, -OH, -CN, NH<sub>2</sub>, -NH(C<sub>1</sub> to C<sub>6</sub> alkyl), or -N(C<sub>1</sub> to C<sub>6</sub> alkyl)<sub>2</sub>;

R<sup>A</sup> is H, C<sub>1</sub> to C<sub>3</sub> alkyl, substituted C<sub>1</sub> to C<sub>3</sub> alkyl, aryl, substituted aryl, C<sub>1</sub> to C<sub>3</sub> alkoxy, substituted C<sub>1</sub> to C<sub>3</sub> alkoxy, C<sub>1</sub> to C<sub>3</sub> aminoalkyl, substituted C<sub>1</sub> to C<sub>3</sub> aminoalkyl,

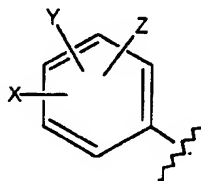
R<sup>B</sup> is H, C<sub>1</sub> to C<sub>3</sub> alkyl, substituted C<sub>1</sub> to C<sub>3</sub> alkyl,

10 R<sup>3</sup> is H, OH, NH<sub>2</sub>, C<sub>1</sub> to C<sub>6</sub> alkyl, substituted C<sub>1</sub> to C<sub>6</sub> alkyl, C<sub>3</sub> to C<sub>6</sub> alkenyl, substituted C<sub>1</sub> to C<sub>6</sub> alkenyl, alkynyl, or substituted alkynyl, COR<sup>C</sup>,

R<sup>C</sup> is H, C<sub>1</sub> to C<sub>3</sub> alkyl, substituted C<sub>1</sub> to C<sub>3</sub> alkyl, aryl, substituted aryl, C<sub>1</sub> to C<sub>3</sub> alkoxy, substituted C<sub>1</sub> to C<sub>3</sub> alkoxy, C<sub>1</sub> to C<sub>3</sub> aminoalkyl, substituted C<sub>1</sub> to C<sub>3</sub> aminoalkyl,

15 R<sup>4</sup> is H, halogen, CN, NO<sub>2</sub>, C<sub>1</sub> to C<sub>6</sub> alkyl, substituted C<sub>1</sub> to C<sub>6</sub> alkyl, alkynyl, or substituted alkynyl, C<sub>1</sub> to C<sub>6</sub> alkoxy, substituted C<sub>1</sub> to C<sub>6</sub> alkoxy, amino, C<sub>1</sub> to C<sub>6</sub> aminoalkyl, substituted C<sub>1</sub> to C<sub>6</sub> aminoalkyl,

R<sup>5</sup> is a trisubstituted benzene ring containing the substituents X, Y and Z as shown below,



20

X is taken from the group including halogen, CN, C<sub>1</sub> to C<sub>3</sub> alkyl, substituted C<sub>1</sub> to C<sub>3</sub> alkyl, alkynyl, or substituted alkynyl, C<sub>1</sub> to C<sub>3</sub> alkoxy, substituted C<sub>1</sub> to C<sub>3</sub> alkoxy, C<sub>1</sub> to C<sub>3</sub> thioalkoxy, substituted C<sub>1</sub> to C<sub>3</sub> thioalkoxy, amino, C<sub>1</sub> to C<sub>3</sub> aminoalkyl, substituted C<sub>1</sub> to C<sub>3</sub> aminoalkyl, NO<sub>2</sub>, C<sub>1</sub> to C<sub>3</sub> perfluoroalkyl, 5 or 6

- 10 -

membered heterocyclic ring containing 1 to 3 heteroatoms,  $\text{COR}^D$ ,  $\text{OCOR}^D$ , or  $\text{NR}^E\text{COR}^D$ ;

$\text{R}^D$  is H,  $\text{C}_1$  to  $\text{C}_3$  alkyl, substituted  $\text{C}_1$  to  $\text{C}_3$  alkyl, aryl, substituted aryl,  $\text{C}_1$  to  $\text{C}_3$  alkoxy, substituted  $\text{C}_1$  to  $\text{C}_3$  alkoxy,  $\text{C}_1$  to  $\text{C}_3$  aminoalkyl, or substituted  $\text{C}_1$  to  $\text{C}_3$  aminoalkyl;

$\text{R}^E$  is H,  $\text{C}_1$  to  $\text{C}_3$  alkyl, substituted  $\text{C}_1$  to  $\text{C}_3$  alkyl;

Y and Z are independent substituents taken from the group including H, halogen, CN,  $\text{NO}_2$ , amino, aminoalkyl,  $\text{C}_1$  to  $\text{C}_3$  alkoxy,  $\text{C}_1$  to  $\text{C}_3$  alkyl, or  $\text{C}_1$  to  $\text{C}_3$  thioalkoxy;

or

$\text{R}^5$  is a five or six membered ring with 1, 2, or 3 heteroatoms from the group including O, S, SO,  $\text{SO}_2$  or  $\text{NR}^6$  and containing one or two independent substituents from the group including H, halogen, CN,  $\text{NO}_2$ , amino, and  $\text{C}_1$  to  $\text{C}_3$  alkyl,  $\text{C}_1$  to  $\text{C}_3$  alkoxy,  $\text{C}_1$  to  $\text{C}_3$  aminoalkyl,  $\text{COR}^F$ , or  $\text{NR}^G\text{COR}^F$ ;

$\text{R}^F$  is H,  $\text{C}_1$  to  $\text{C}_3$  alkyl, substituted  $\text{C}_1$  to  $\text{C}_3$  alkyl, aryl, substituted aryl,  $\text{C}_1$  to  $\text{C}_3$  alkoxy, substituted  $\text{C}_1$  to  $\text{C}_3$  alkoxy,  $\text{C}_1$  to  $\text{C}_3$  aminoalkyl, or substituted  $\text{C}_1$  to  $\text{C}_3$  aminoalkyl;

$\text{R}^G$  is H,  $\text{C}_1$  to  $\text{C}_3$  alkyl, or substituted  $\text{C}_1$  to  $\text{C}_3$  alkyl;

$\text{R}^6$  is H or  $\text{C}_1$  to  $\text{C}_3$  alkyl;

$\text{G}_1$  is O,  $\text{NR}_7$ , or  $\text{CR}_7\text{R}_8$ ;

$\text{G}_2$  is CO, CS, or  $\text{CR}_7\text{R}_8$ ;

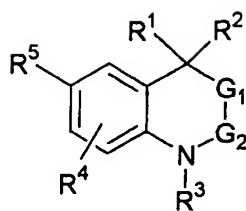
provided that when  $\text{G}_1$  is O,  $\text{G}_2$  is  $\text{CR}_7\text{R}_8$ , and  $\text{G}_1$  and  $\text{G}_2$  cannot both be  $\text{CR}_7\text{R}_8$ ;

$\text{R}_7$  and  $\text{R}_8$  are independent substituents selected from H or an optionally substituted alkyl, aryl, or heterocyclic moiety;

or pharmaceutically acceptable salt thereof.

Preferred compounds are those of Formula I

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I

wherein:

$R^1$  is H,  $C_1$  to  $C_6$  alkyl, substituted  $C_1$  to  $C_6$  alkyl,  $C_3$  to  $C_8$  cycloalkyl, substituted  $C_3$  to  $C_8$  cycloalkyl, aryl, substituted aryl, heterocyclic, substituted heterocyclic,  $COR^A$ , or  $NR^B COR^A$ ;

$R^2$  is H,  $C_1$  to  $C_6$  alkyl, substituted  $C_1$  to  $C_6$  alkyl,  $C_2$  to  $C_6$  alkenyl, substituted  $C_2$  to  $C_6$  alkenyl,  $C_3$  to  $C_8$  cycloalkyl, substituted  $C_3$  to  $C_8$  cycloalkyl, aryl, substituted aryl, heterocyclic, substituted heterocyclic,  $COR^A$ , or  $NR^B COR^A$ ;

or

$R^1$  and  $R^2$  are fused to form an optionally substituted 3 to 8 membered spirocyclic alkyl, alkenyl or heterocyclic ring containing one to three heteroatoms from the group including O, S and N, as described above;

$R^A$  is H,  $C_1$  to  $C_3$  alkyl, substituted  $C_1$  to  $C_3$  alkyl, aryl, substituted aryl,  $C_1$  to  $C_3$  alkoxy, substituted  $C_1$  to  $C_3$  alkoxy,  $C_1$  to  $C_3$  aminoalkyl, or substituted  $C_1$  to  $C_3$  aminoalkyl;

$R^B$  is H,  $C_1$  to  $C_3$  alkyl, or substituted  $C_1$  to  $C_3$  alkyl,

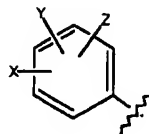
$R^3$  is H, OH,  $NH_2$ ,  $C_1$  to  $C_6$  alkyl, substituted  $C_1$  to  $C_6$  alkyl,  $C_3$  to  $C_6$  alkenyl, substituted  $C_1$  to  $C_6$  alkenyl, alkynyl, or substituted alkynyl,  $COR^C$ ;

$R^C$  is H,  $C_1$  to  $C_4$  alkyl, substituted  $C_1$  to  $C_4$  alkyl, aryl, substituted aryl,  $C_1$  to  $C_4$  alkoxy, substituted  $C_1$  to  $C_4$  alkoxy,  $C_1$  to  $C_4$  aminoalkyl, or substituted  $C_1$  to  $C_4$  aminoalkyl;

$R^4$  is H, halogen, CN,  $NO_2$ ,  $C_1$  to  $C_6$  alkyl, substituted  $C_1$  to  $C_6$  alkyl,  $C_1$  to  $C_6$  alkoxy, substituted  $C_1$  to  $C_6$  alkoxy, amino,  $C_1$  to  $C_6$  aminoalkyl, substituted  $C_1$  to  $C_6$  aminoalkyl,

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$R^5$  is a trisubstituted benzene ring containing the substituents X, Y and Z as shown below:

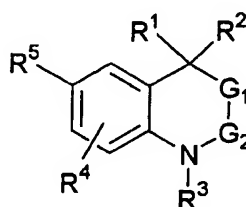


- 5 X is selected from halogen, CN,  $C_1$  to  $C_3$  alkyl, substituted  $C_1$  to  $C_3$  alkyl,  $C_1$  to  $C_3$  alkoxy, substituted  $C_1$  to  $C_3$  alkoxy,  $C_1$  to  $C_3$  thioalkoxy, substituted  $C_1$  to  $C_3$  thioalkoxy, amino,  $C_1$  to  $C_3$  aminoalkyl, substituted  $C_1$  to  $C_3$  aminoalkyl,  $NO_2$ ,  $C_1$  to  $C_3$  perfluoroalkyl, 5 membered heterocyclic ring containing 1 to 3 heteroatoms,  $COR^D$ ,  $OCOR^D$ , or  $NR^E COR^D$ ;
- 10  $R^D$  is H,  $C_1$  to  $C_3$  alkyl, substituted  $C_1$  to  $C_3$  alkyl, aryl, substituted aryl,  $C_1$  to  $C_3$  alkoxy, substituted  $C_1$  to  $C_3$  alkoxy,  $C_1$  to  $C_3$  aminoalkyl, or substituted  $C_1$  to  $C_3$  aminoalkyl;
- $R^E$  is H,  $C_1$  to  $C_3$  alkyl, or substituted  $C_1$  to  $C_3$  alkyl;
- Y and Z are independent substituents taken from the group including H,
- 15 halogen, CN,  $NO_2$ ,  $C_1$  to  $C_3$  alkoxy,  $C_1$  to  $C_3$  alkyl, or  $C_1$  to  $C_3$  thioalkoxy;
- or
- $R^5$  is a five or six membered ring with 1, 2, or 3 heteroatoms from the group including O, S, SO,  $SO_2$  or  $NR^6$  and containing one or two independent substituents from the group including H, halogen, CN,  $NO_2$ , amino, and  $C_1$  to  $C_3$  alkyl,  $C_1$  to  $C_3$
- 20 alkoxy.
- $R^6$  is H, or  $C_1$  to  $C_3$  alkyl.
- $G_1$  is O,  $NR_7$ , or  $CR_7R_8$
- $G_2$  is CO, CS, or  $CR_7R_8$ , with the proviso that when  $G_1$  is O,  $G_2$  is  $CR_7R_8$ , and  $G_1$  and  $G_2$  cannot both be  $CR_7R_8$ ;
- 25 wherein  $R_7$  and  $R_8$  are independent substituent selected from H, alkyl, substituted alkyl, aryl, substituted aryl, heterocyclic, or substituted heterocyclic

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or a pharmaceutically acceptable salt thereof.

Still, more preferred compounds are those of Formula I



I

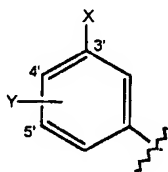
5 wherein:

$R^1 = R^2$  and are selected from  $C_1$  to  $C_3$  alkyl, substituted  $C_1$  to  $C_3$  alkyl, or spirocyclic alkyl constructed by fusing  $R^1$  and  $R^2$  to form a 3 to 6 membered spirocyclic ring;

10  $R^3$  is H, OH,  $NH_2$ ,  $C_1$  to  $C_6$  alkyl, substituted  $C_1$  to  $C_6$  alkyl,  $-COH$ ,  $-CO(C_1$  to  $C_4$  alkyl) or  $-CO(C_1$  to  $C_4$  alkoxy);

$R^4$  is H, halogen,  $NO_2$ ,  $C_1$  to  $C_3$  alkyl, substituted  $C_1$  to  $C_3$  alkyl,

$R^5$  is a disubstituted benzene ring containing the substituents X, and Y as shown below



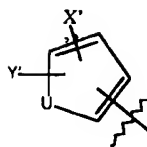
15 X is taken from the group including halogen, CN,  $C_1$  to  $C_3$  alkoxy,  $C_1$  to  $C_3$  alkyl,  $NO_2$ ,  $C_1$  to  $C_3$  perfluoroalkyl, 5 membered heterocyclic ring containing 1 to 3 heteroatoms,  $C_1$  to  $C_3$  thioalkoxy,

Y is a substituent on the 4' or 5' position from the group including H, halogen, CN,  $NO_2$ ,  $C_1$  to  $C_3$  alkoxy,  $C_1$  to  $C_4$  alkyl,  $C_1$  to  $C_3$  thioalkoxy

20 or

$R^5$  is a five membered ring with the structure shown below

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U is O, S, or NR<sup>6</sup>,

R<sup>6</sup> is H, or C<sub>1</sub> to C<sub>3</sub> alkyl, C<sub>1</sub> to C<sub>4</sub> CO<sub>2</sub>alkyl,

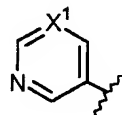
5 X' is from the group including halogen, CN, NO<sub>2</sub>, C<sub>1</sub> to C<sub>3</sub> alkyl and C<sub>1</sub> to C<sub>3</sub> alkoxy;

Y' is from the group including H and C<sub>1</sub> to C<sub>4</sub> alkyl

or

X<sup>5</sup> is a six membered ring with the structure shown

10



X<sup>1</sup> is N or CX<sup>2</sup>,

X<sup>2</sup> is halogen, CN, alkoxy, or NO<sub>2</sub>,

15 G<sub>1</sub> is O, NR<sub>7</sub>, or CR<sub>7</sub>R<sub>8</sub>

G<sub>2</sub> is CO, CS, or CR<sub>7</sub>R<sub>8</sub>

provided that when G<sub>1</sub> is O, G<sub>2</sub> is CR<sub>7</sub>R<sub>8</sub>, and G<sub>1</sub> and G<sub>2</sub> cannot both be CR<sub>7</sub>R<sub>8</sub>;

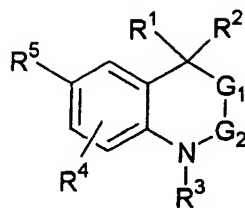
wherein R<sub>7</sub> and R<sub>8</sub> are independent substituents selected from H, alkyl, substituted alkyl, aryl, substituted aryl, heterocyclic, or substituted heterocyclic;

20 or pharmaceutically acceptable salt thereof

Still, even more preferred compounds are those of Formula I



- 15 -



I

wherein:

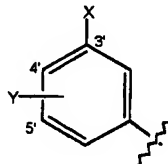
$R^1 = R^2$  and are selected from the group which includes  $\text{CH}_3$  and spirocyclic alkyl constructed by fusing  $R^1$  and  $R^2$  to form a 6 membered spirocyclic ring,

5  $R^3$  is H, OH,  $\text{NH}_2$ ,  $\text{CH}_3$ , substituted methyl,  $\text{COR}^C$ ,

$R^C$  is H,  $\text{C}_1$  to  $\text{C}_3$  alkyl,  $\text{C}_1$  to  $\text{C}_4$  alkoxy,

$R^4$  is H, halogen,  $\text{C}_1$  to  $\text{C}_3$  alkyl,

$R^5$  is a disubstituted benzene ring containing the substituents X, and Y as shown below



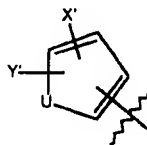
10

X is taken from the group including halogen, CN, methoxy,  $\text{NO}_2$ , 2-thiazole,

Y is a substituent on the 4' or 5' position from the group including H and F,

or

$R^5$  is a five membered ring with the structure shown below



15

U is O, S, or NH,

$X'$  is from the group including halogen, CN,  $\text{NO}_2$ ,

$Y'$  is from the group including H and  $\text{C}_1$  to  $\text{C}_4$  alkyl

$G_1$  is O,  $\text{NR}_7$ , or  $\text{CR}_7\text{R}_8$

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$G_2$  is CO, CS, or  $CR_7R_8$

provided that when  $G_1$  is O,  $G_2$  is  $CR_7R_8$ , and  $G_1$  and  $G_2$  cannot both be  $CR_7R_8$ ;

$R_7$  and  $R_8$  are independent substituent selected from H, alkyl, substituted alkyl,  
5 aryl, substituted aryl, heterocyclic, or substituted heterocyclic;  
and pharmaceutically acceptable salts thereof.

The compounds of this invention may contain an asymmetric carbon atom and some of the compounds of this invention may contain one or more asymmetric centers and may thus give rise to optical isomers and diastereomers. While shown without  
10 respect to stereochemistry in Formula I, the present invention includes such optical isomers and diastereomers; as well as the racemic and resolved, enantiomerically pure R and S stereoisomers; as well as other mixtures of the R and S stereoisomers and pharmaceutically acceptable salts thereof. It will be understood by one skilled in the art that the number of substituents listed for spirocyclic or heterospirocyclic rings  
15 formed by fusing  $R_1$  and  $R_2$  will be determined by the size of the spirocyclic ring.

The term "alkyl" is used herein to refer to both straight- and branched-chain saturated aliphatic hydrocarbon groups having one to eight carbon atoms; "alkenyl" is intended to include both straight- and branched-chain alkyl group with at least one carbon-carbon double bond and two to eight carbon atoms; "alkynyl" group is  
20 intended to cover both straight- and branched-chain alkyl group with at least one carbon-carbon triple bond and two to eight carbon atoms.

The terms "substituted alkyl", "substituted alkenyl", and "substituted alkynyl" refer to alkyl, alkenyl, and alkynyl as just described having one or more substituents from the group including halogen, CN, OH,  $NO_2$ , amino, aryl, heterocyclic, substituted  
25 aryl, substituted heterocyclic, alkoxy, aryloxy, substituted alkyloxy, alkylcarbonyl, alkylcarboxy, alkylamino, arylthio. These substituents may be attached to any carbon of alkyl, alkenyl, or alkynyl group provided that the attachment constitutes a stable chemical moiety.

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The term "aryl" is used herein to refer to an aromatic system which may be a single ring or multiple aromatic rings fused or linked together as such that at least one part of the fused or linked rings forms the conjugated aromatic system. The aryl groups include but not limited to phenyl, naphthyl, biphenyl, anthryl, tetrahydronaphthyl, phenanthryl.

The term "substituted aryl" refers to aryl as just defined having one to four substituents from the group including halogen, CN, OH, NO<sub>2</sub>, amino, alkyl, cycloalkyl, alkenyl, alkynyl, alkoxy, aryloxy, substituted alkyloxy, alkylcarbonyl, alkylcarboxy, alkylamino, or arylthio.

The term "heterocyclic" is used herein to describe a stable 4- to 7-membered monocyclic or a stable multicyclic heterocyclic ring which is saturated, partially unsaturated, or unsaturated, and which consists of carbon atoms and from one to four heteroatoms selected from the group including N, O, and S atoms. The N and S atoms may be oxidized. The heterocyclic ring also includes any multicyclic ring in which any of above defined heterocyclic rings is fused to an aryl ring. The heterocyclic ring may be attached at any heteroatom or carbon atom provided the resultant structure is chemically stable. Such heterocyclic groups include, for example, tetrahydrofuran, piperidinyl, piperazinyl, 2-oxopiperidinyl, azepinyl, pyrrolidinyl, imidazolyl, pyridyl, pyrazinyl, pyrimidinyl, pyridazinyl, oxazolyl, isoxazolyl, morpholinyl, indolyl, quinolinyl, thienyl, furyl, benzofuranyl, benzothienyl, thiamorpholinyl, thiamorpholinyl sulfoxide, and isoquinolinyl.

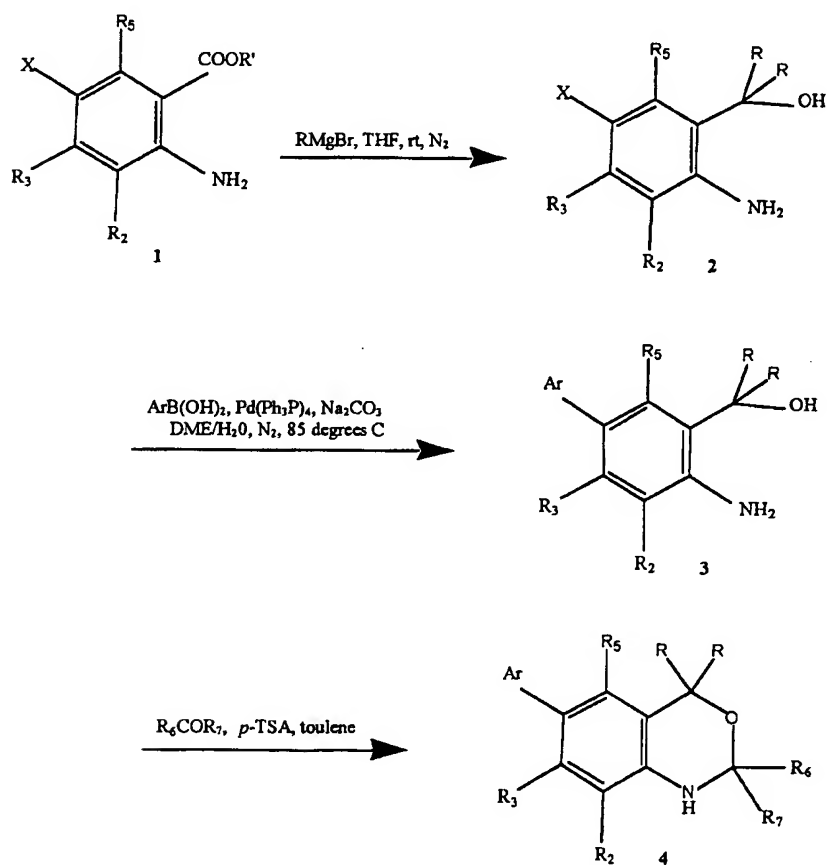
The term "substituted heterocyclic" is used herein to describe the heterocyclic just defined having one to four substituents selected from the group which includes halogen, CN, OH, NO<sub>2</sub>, amino, alkyl, substituted alkyl, cycloalkyl, alkenyl, substituted alkenyl, alkynyl, alkoxy, aryloxy, substituted alkyloxy, alkylcarbonyl, alkylcarboxy, alkylamino, or arylthio. The term "alkoxy" refers to the OR group, where R is alkyl or substituted alkyl. The term "aryloxy" is used herein to indicate the OR group, where R is aryl or substituted aryl. The term "alkylcarbonyl" refers to the RCO group, where R

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is alkyl or substituted alkyl. The term "alkylcarboxy" refers to the COOR group, where R is alkyl or substituted alkyl. The term "aminoalkyl" refers to both secondary and tertiary amines wherein the alkyl or substituted alkyl groups, containing one to eight carbon atoms, which may be either same or different and the point of attachment is on the nitrogen atom. The term "halogen" refers to Cl, Br, F, and I element.

The compounds of this invention can be prepared following the Schemes illustrated below:

Scheme I



As demonstrated in Scheme I, the compounds of this invention are generally prepared by employing the suitable coupling reaction as a key step. An appropriately substituted ortho-amino benzoic acid or its derivatives such as ethyl ester (X = Br, I,

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Cl, or a latent coupling precursor such as alkoxy group which can be converted into OTf group suitable in the coupling reaction) was treated with a suitable organo metallic reagent, e.g. Grignard reagent, in appropriate nonprotic solvents which include but are not limited to THF or ether to give ortho-amino carbinol 2 under an inert atmosphere such as argon or nitrogen at -78 °C to room temperature. The arylation of amino carbinol 2 to yield 3 can be effected by various coupling reactions including Suzuki, Stille reactions. These reactions are commonly performed in the presence of a transition metallic catalyst, e.g., palladium or nickel complex often with phosphino ligands, e.g., Ph<sub>3</sub>P, dppf, dppe or a catalyst such as palladium acetate.

Under this catalytic condition, an appropriately substituted nucleophilic reagent, e.g., aryl boronic acid, arylstannane, or aryl zinc compound, is coupled with amino carbinol 2 to give 3. If a base is needed in the reaction, the commonly used bases include but are not limited to sodium bicarbonate, sodium carbonate, potassium phosphate, barium carbonate, cesium fluoride, or potassium acetate. The most commonly used solvents in these reactions include benzene, DMF, isopropanol, ethanol, DME, ether, acetone, or a mixture of above solvent and water. The coupling reaction is generally executed under an inert atmosphere such as nitrogen or argon at temperatures ranging from room temperature to 95 °C. The compounds of this invention 4 can be effected by treatment of amino carbinol 3 with an appropriate ketone in the presence of an suitable acid catalyst such as *p*-toluenesulfonic acid in a suitable solvent such as toluene, benzene under an inert atmosphere such as argon or nitrogen at room temperature to reflux.

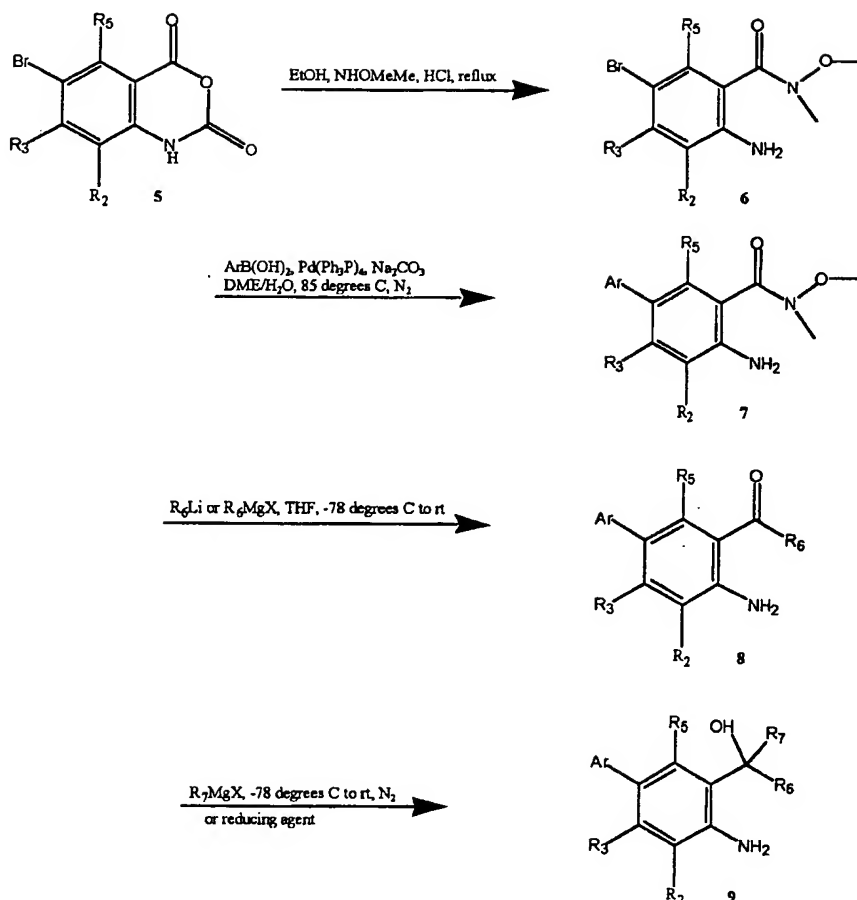
Scheme II describes the procedure to prepare benzoxazines bearing two different substituents at position-4. The Weinreb amide 6 can be prepared from an appropriately substituted isatoic anhydride 5 when treated with *N*-, *O*-dimethylhydroxyl-amine hydrochloride salt in a protic solvent such as ethanol, or isopropanol at reflux under an inert atmosphere such as argon or nitrogen. Coupling of amide 6 with an aryl electrophile such as aryl boronic acid or arylstannane to give 7

- 20 -

can be effected by employing a typical coupling reaction such as Suzuki, Stille coupling procedure in a similar fashion as described for the preparation of compound 3. Treatment of Weinreb amide 7 with organo metallic compounds, e.g., alkyllithium, alkynyllithium, aryllithium, or their Grignard counterpart in a nonprotic solvent such as THF or ether under an inert atmosphere such as argon or nitrogen at -78 ° to room temperature affords amino ketone 8. Conversion of ketone 8 to carbinol 9 can be effected by treatment of 8 with an organo metallic reagent such as alkyl, alkynyl, or aryl Grignard reagent in a nonprotic solvent such as THF or ether under an inert atmosphere such as argon or nitrogen at -78 °C to room temperature. Conversion of ketone 8 to carbinol 9 can also be effected by reduction of ketone group of 8 to the carbinol moiety of 9 using an appropriate reducing reagent such as lithium aluminum hydride, sodium borohydride in a suitable solvent such as THF, ether, or anhydrous alcohol under an inert atmosphere in the temperature ranging from 0 °C to the boiling point of the solvent. Further conversion of 9 to the compounds of this invention can be effected as described in scheme I for the preparation of compound 4.

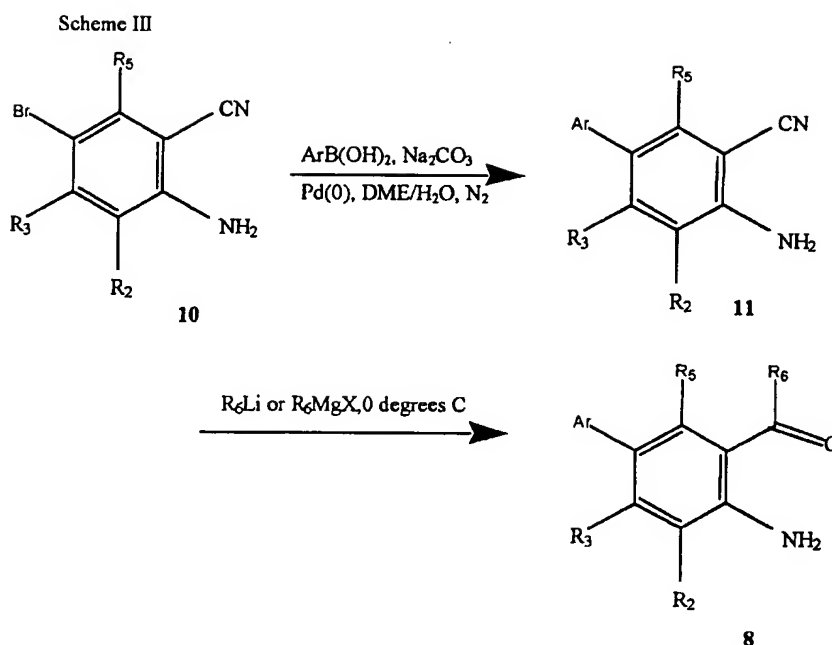
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Scheme II



Alternatively, ortho-amino ketone **8** can be prepared by treatment of ortho-amino benzonitrile **11** with an organo metallic compound such as organo lithium reagent or Grignard reagent in a suitable solvent such as THF or ether under an inert atmosphere such as argon or nitrogen at temperatures ranging from -78 °C to room temperature as illustrated in Scheme III. Benzonitrile **11** can be readily prepared from an appropriately substituted benzonitrile such as bromobenzonitrile **10** using a suitable coupling reaction such as Stille or Suzuki protocol carried out in a similar fashion as described for the preparation of the Weinreb amide **7**.

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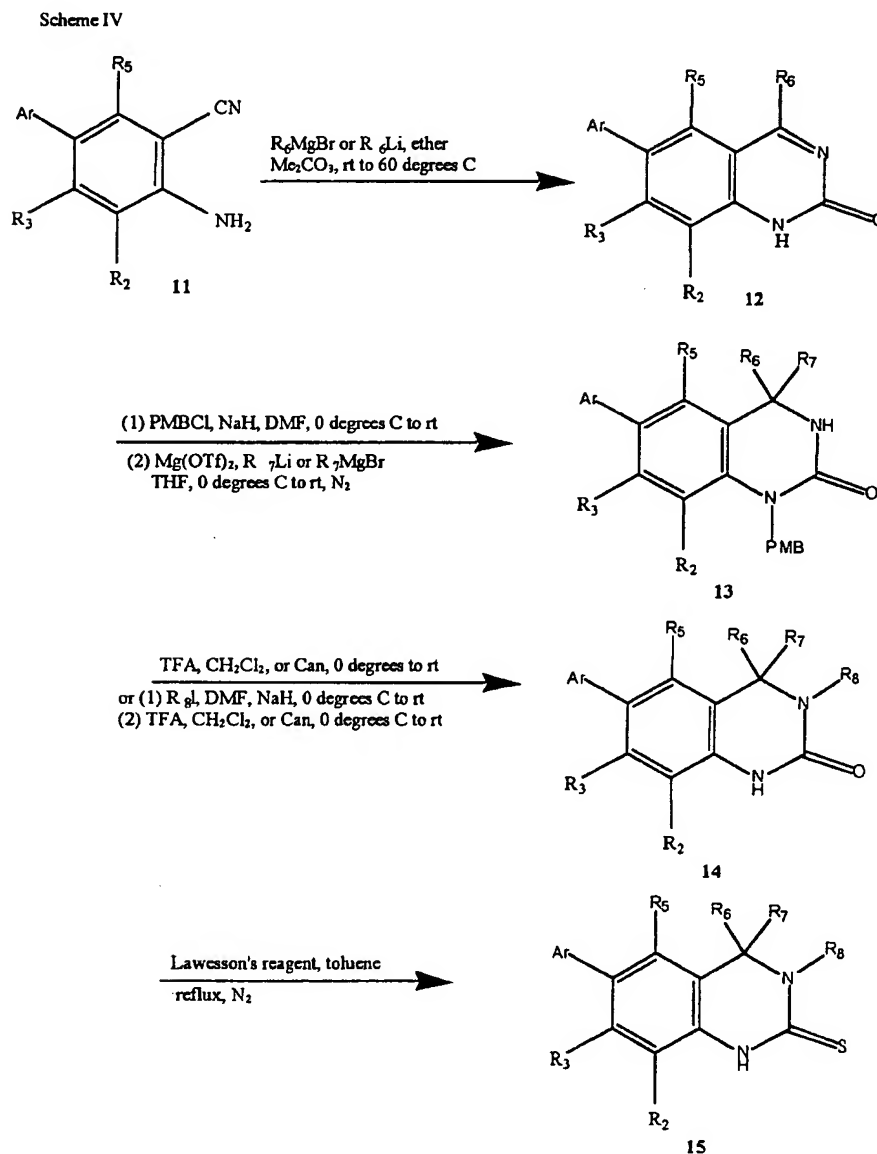
- Scheme IV illustrates the synthesis of 3,4-dihydroquinazolin-2-ones. The substituted 2-aminobenzonitrile **11** is treated with an organo metallic compound such as an organo lithium or Grignard reagent in a nonprotic solvent such as THF or ether under an inert atmosphere such argon or nitrogen at -78 °C to room temperature to produce an imino intermediate which is reacted with a suitable carbonate such as diethyl carbonate or dimethyl carbonate *in situ* at 0 °C to 60 °C to give quinazolin-2-ones **12**. Protection of quinazolin-2-ones **12** with a suitable protective group such as a *para*-methoxybenzyl moiety can be effected by treating **12** with a suitable base such as potassium hydride, potassium *t*-butoxide, or sodium hydride followed by addition of a protective reagent such as *para*-methoxybenzyl chloride in an appropriate solvent such as DMF, or a mixture of solvents such as THF and DMF under an inert atmosphere such as nitrogen or argon at 0 °C to room temperature. The Michael addition of a suitable organo metallic compound such as organo lithium or Grignard reagent to the



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protected quinazolin-2-ones **12** to give **13** can be accomplished in the presence of a suitable Lewis acid such as magnesium triflate in a nonprotic solvent such as THF or ether under an inert atmosphere such as argon or nitrogen at 0 °C to room temperature.

5



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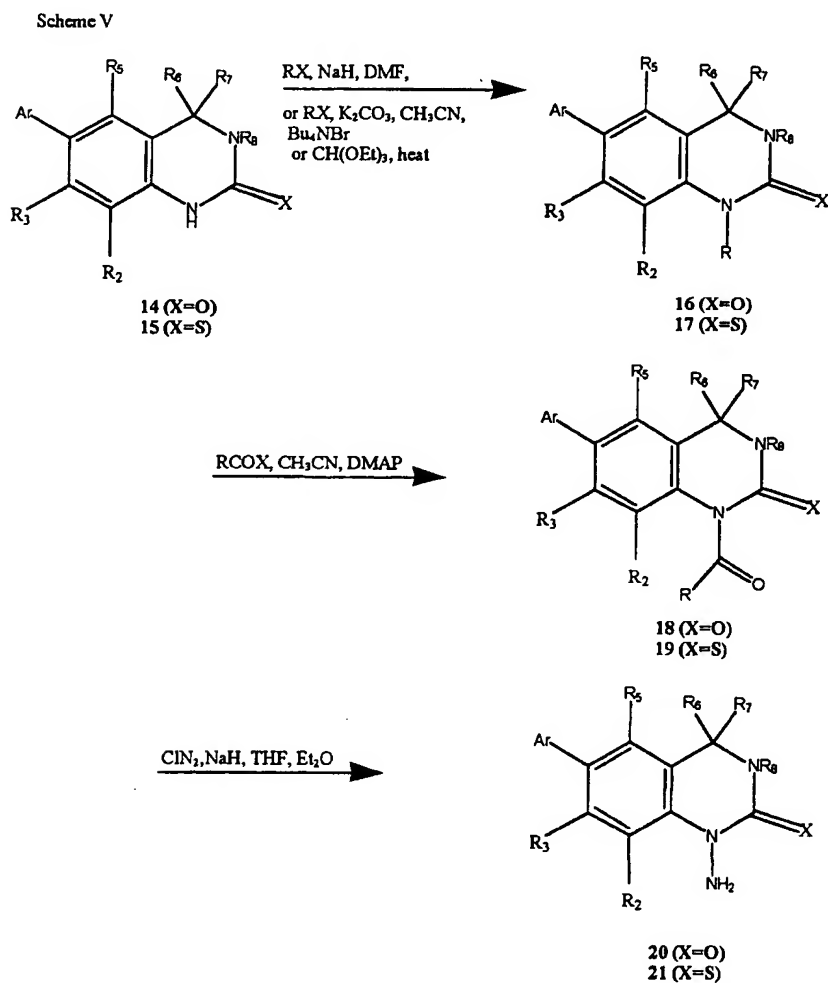
Removal of the protective group can be effected by treating **13** with a suitable deprotecting reagent, e.g. for the *para*-methoxybenzyl protective group it can be removed by treatment of **13** with protic acid such as TFA or with Ceric ammonium nitrate in a suitable solvent such as methylene chloride at 0 °C to room temperature  
5 under an inert atmosphere such as argon or nitrogen. Prior to the removal of protective group, the alkylation of 3-nitrogen can be achieved by treating **13** with an appropriate base such as sodium hydride, potassium hydride, or potassium *t*-butoxide in a suitable solvent such as DMF followed by quenching the reaction solution with an organo iodide or an organo triflate such as iodomethane under an inert atmosphere  
10 such as argon or nitrogen at 0 °C to room temperature. The compounds of the present invention **14** can be prepared when the protective group is removed with a suitable reagent, e.g. for the *para*-methoxybenzyl protective group it can be removed by treatment of **13** with protic acid such as TFA or with Ceric ammonium nitrate in a suitable solvent such as methylene chloride at 0 °C to room temperature under an inert  
15 atmosphere such as argon or nitrogen.

The conversion of compounds **14** to 3,4-dihydroquinazolin-2-thiones **15** can be accomplished by treatment of **14** with a suitable sulfur reagent such as Lawesson's reagent in a nonprotic solvent such as *o*-xylene, chlorobenzene, or toluene under an inert atmosphere such as argon or nitrogen at reflux.

20 As illustrated in scheme V, the compounds **14** or **15** can be further derivatized at position-1 *via* numerous approaches leading to a variety of the novel derivatives including 1-alkyl, substituted 1-alkyl, 1-carbonyl, substituted 1-carbonyl, 1-carboxy, substituted 1-carboxy derivatives. For example, alkyl or substituted alkyl derivatives  
25 **16** or **17** can be formed by treatment of carbamate **14** or **15** with a suitable base such as sodium hydride in a suitable solvent such as DMF under an inert atmosphere such as argon or nitrogen followed by addition of an appropriate electrophile such as alkyl or substituted alkyl bromide, iodide, or triflate. Such transformation of **14** or **15** to **16** or **17** at position-1 can also be effected using biphasic condition as indicated in scheme V

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in which alkylation is executed using a biphasic catalyst such as tributylammonium bromide in a suitable solvent such as acetonitrile. A further example of such modification in position-1 includes but not limited to the one depicted in scheme V in that heating of 14 or 15 with triethyl orthoformate affords 1-substituted derivatives of compound 14 or 15.



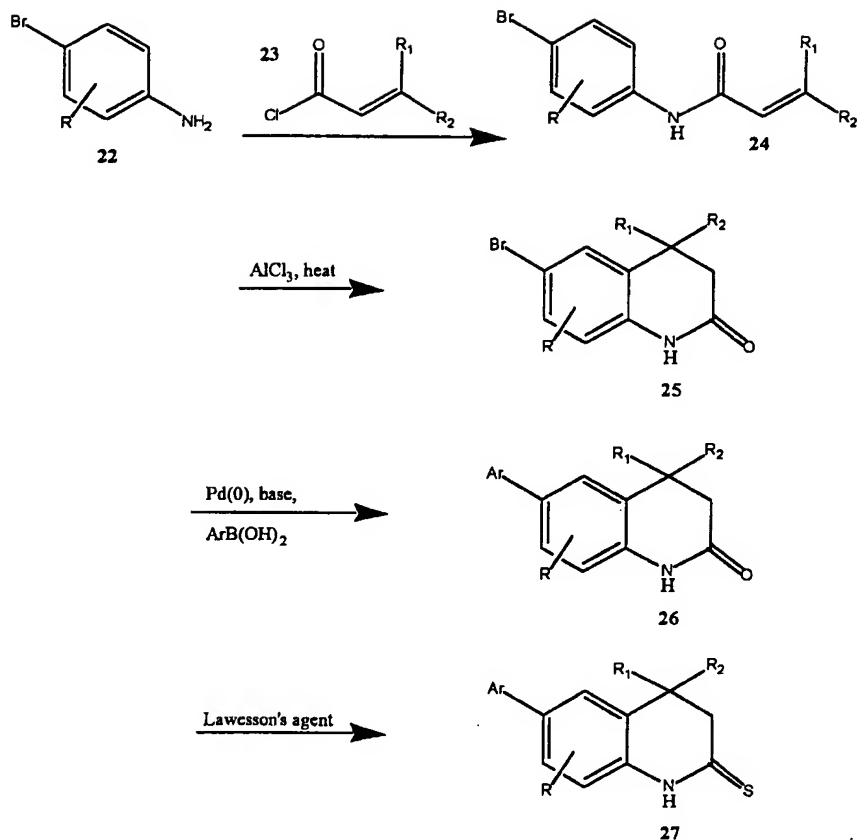
- 26 -

The acylation or carboxylation of the compound **14** or **15** at position-1 to give compound **18** or **19** can be readily effected by treatment of **14** or **15** with a suitable acylating or carboxylating reagent such as di-*t*-butyl dicarbonate in the presence of a suitable basic catalyst such as DMAP in a suitable solvent such as acetonitrile under an inert atmosphere such as argon or nitrogen. The amination of position-1 of compound **14** or **15** to give compounds **20** and **21** can be furnished using a suitable aminating reagent such as chloroamine in the presence of a suitable base such as sodium hydride in a suitable solvent such as THF or diethyl ether following the literature procedure (Metlesics et al. *J. Org. Chem.* **30**, 1311(1965)).

According to scheme VI an appropriate aniline such as 4-bromoaniline **22**, is reacted in the presence of a base in a suitable nonprotic solvent with an acryloyl chloride **23** to form the amide **24**. The base is preferably a strong base such as sodium hydride or sodium or potassium hexamethyldisilylamide, utilizing THF as the solvent under an inert atmosphere (nitrogen or argon) from 0°C up to the reflux temperature of the solvent.

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Scheme VI



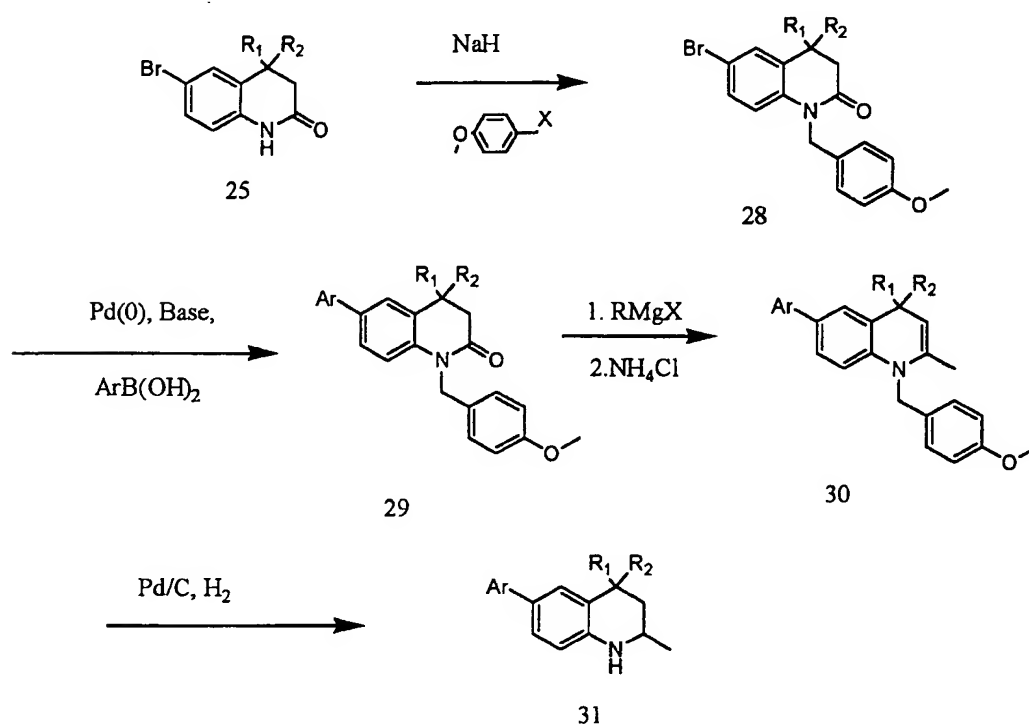
Reaction of the amide 24 under strongly acidic conditions, sulfuric acid, borontrifluoride etherate, or preferably aluminum chloride either as a melt, or in an inert solvent (dichlorobenzenes) under an inert atmosphere (nitrogen or argon) from 0 °C up to the reflux temperature, of the solvent then provides the cyclic amide 25. Subsequent reaction of compound 25 with an aryl or heteroaryl boronic acid, boronic acid anhydride or trialkyl stannane then provides access to the desired biaryl compound 26. The reaction can be carried out in a solvent such as acetone, ethanol, benzene, toluene or THF, under an inert atmosphere (nitrogen or argon) from 0 °C up to the reflux temperature of the solvent, in the presence of a palladium catalyst such as

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tetrakis(triphenylphosphine) palladium (0) or palladium acetate and may require an additive such as sodium carbonate, cesium fluoride or potassium phosphate.

The conversion of compounds **26** to thioamide **27** can be accomplished by treatment of **26** with a suitable sulfur reagent such as Lawesson's reagent in a nonprotic solvent such as o-xylene, chlorobenzene, or toluene under an inert atmosphere such as argon or nitrogen at reflux.

Scheme VII

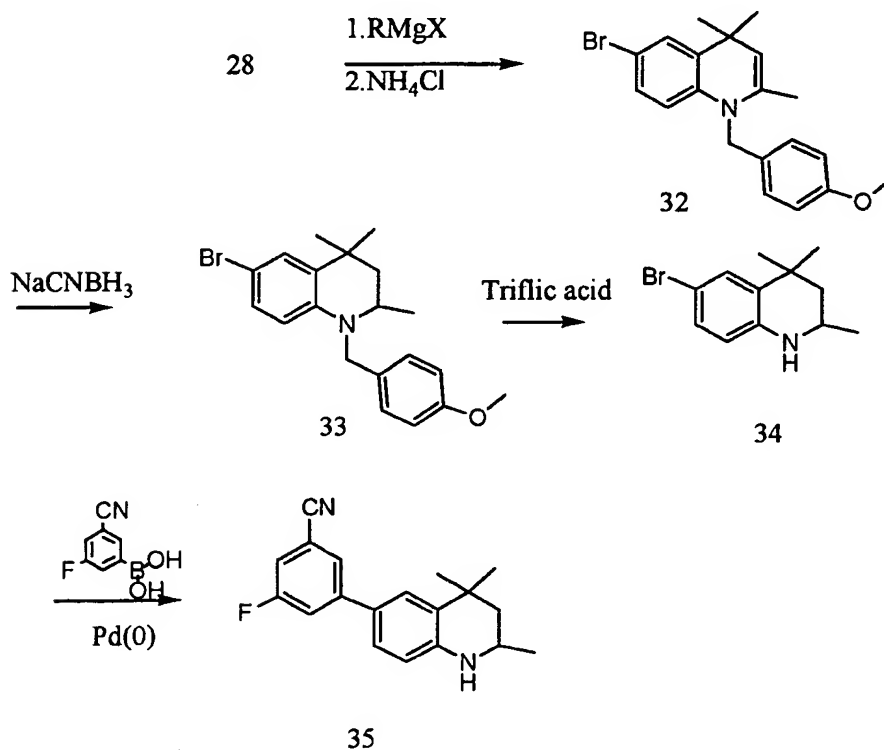


According to scheme VII, an appropriate cyclic amide such as **25**, is allowed to react with NaH in THF to form the anion species and then a benzyl halide is added to convert the starting material to the N-protected amide product, **28**. Reaction of **28** with an aryl boronic acid in the presence of a palladium catalyst such as tetrakis(triphenylphosphine)palladium(0) or palladium acetate permits a coupling of the two aromatic species to yield **29**. The reaction is normally carried out under biphasic

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conditions. That is, water is often employed along with an appropriate organic solvent, such as toluene or DMF. The palladium catalyst is typically added last and the reaction mixture is refluxed in the presence of an inert gas such as nitrogen. The product is treated with a Grignard Reagent, an alkyl magnesium halide, in THF followed by the addition of ammonium chloride solution to afford the enamine derivative **30**. The reduction of the double bond in **30** and removal of the protecting group can be accomplished in a single step by catalytic reduction in a Parr Hydrogenation Apparatus using palladium on charcoal to form the target compound **31**.

Scheme VIII



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3-Fluoro-5-(2,4,4-trimethyl-1,2,3,4-hydro-quinoliny)-benzonitrile (compound 35) can be prepared by Scheme VIII, a process similar to Scheme VII. According to scheme VIII, compound 28 is allowed to react with a Grignard reagent, an alkyl magnesium halide, in THF followed by the addition of ammonium chloride solution to afford the enamine derivative 32. Reduction of the double bond with sodium cyanoborohydride affords the reduced derivative 33. Removal of the protecting group with a strong acid such as triflic or sulfuric acid affords the deprotected compound 34 which can then be coupled with a suitably substituted phenylboronic acid in the presence of a palladium catalyst such as tetrakis(triphenylphosphine)palladium(0) or palladium acetate permits a coupling of the two aromatic species to yield 35. The reaction is normally carried out under biphasic conditions. That is, water is often employed along with an appropriate organic solvent, such as toluene or DMF.

The compounds of the present invention can be used in the form of salts derived from pharmaceutically or physiologically acceptable acids or bases. These salts include, but are not limited to, the following salts with inorganic acids such as hydrochloric acid, sulfuric acid, nitric acid, phosphoric acid and, as the case may be, such organic acids as acetic acid, oxalic acid, succinic acid, and maleic acid. Other salts include salts with alkali metals or alkaline earth metals, such as sodium, potassium, calcium or magnesium in the form of esters, carbamates and other conventional "pro-drug" forms, which, when administered in such form, convert to the active moiety *in vivo*.

This invention includes pharmaceutical compositions and treatments which comprise administering to a mammal a pharmaceutically effective amount of one or more compounds as described above wherein  $G_2$  is  $C=O$  as antagonists of the progesterone receptor. The invention further provides comparable methods and compositions which utilize one or more compounds herein wherein  $G_2$  is  $C=S$  as agonists of the progesterone receptor. Moreover the invention further provides comparable methods and compositions which utilize one or more compounds herein



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wherein  $G_1=O$  and  $G_2=CR_7CR_8$  are agonists of the progesterone receptor and when  $G_1=CR_7CR_8$  and  $G_2=CR_7CR_8$  are agonists of the progesterone receptor.

The progesterone receptor antagonists of this invention, used alone or in combination, can be utilized in methods of contraception and the treatment and/or prevention of benign and malignant neoplastic disease. Specific uses of the compounds and pharmaceutical compositions of invention include the treatment and/or prevention of uterine myometrial fibroids, endometriosis, benign prostatic hypertrophy; carcinomas and adenocarcinomas of the endometrium, ovary, breast, colon, prostate, pituitary, meningioma and other hormone-dependent tumors. Additional uses of the present progesterone receptor antagonists include the synchronization of the estrus in livestock. When used in contraception the progesterone receptor antagonists of the current invention may be used either alone in a continuous administration of between 1 and 500 mg per day, or alternatively used in a different regimen which would entail 2-4 days of treatment with the progesterone receptor antagonist after 21 days of a progestin, in this regimen between 0.1 and 500 mg daily doses of the progestin (e.g. levonorgestrel, trimegestone, gestodene, norethistrone acetate, norgestimate or cyproterone acetate) would be followed by between 0.1 and 500 mg daily doses of the progesterone receptor antagonists of the current invention.

The progesterone receptor antagonists of this invention, used alone or in combination, can also be utilized in methods of treatment and/or prevention of benign and malignant neoplastic disease. Specific uses of the compounds and pharmaceutical compositions of invention include the treatment and/or prevention of uterine myometrial fibroids, endometriosis, benign prostatic hypertrophy; carcinomas and adenocarcinomas of the endometrium, ovary, breast, colon, prostate, pituitary, meningioma and other hormone-dependent tumors. Additional uses of the present progesterone receptor antagonists include the synchronization of the estrus in livestock.

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The progesterone receptor agonists of this invention, used alone or in combination, can be utilized in methods of contraception and the treatment and/or prevention of dysfunctional bleeding, uterine leiomyomata, endometriosis, polycystic ovary syndrome, carcinomas and adenocarcinomas of the endometrium, ovary, breast, colon, prostate. Additional uses of the invention include stimulation of food intake.

When used in contraception the progesterone receptor agonists of the current invention are preferably used in combination or sequentially with an estrogen agonist (e.g. ethinyl estradiol). The preferred dose of the progesterone receptor agonist is between 0.01 and 500 mg per day.

When the compounds are employed for the above utilities, they may be combined with one or more pharmaceutically acceptable carriers or excipients, for example, solvents, diluents and the like, and may be administered orally in such forms as tablets, capsules, dispersible powders, granules, or suspensions containing, for example, from about 0.05 to 5% of suspending agent, syrups containing, for example, from about 10 to 50% of sugar, and elixirs containing, for example, from about 20 to 50% ethanol, and the like, or parenterally in the form of sterile injectable solutions or suspensions containing from about 0.05 to 5% suspending agent in an isotonic medium. Such pharmaceutical preparations may contain, for example, from about 25 to about 90% of the active ingredient in combination with the carrier, more usually between about 5% and 60% by weight.

The effective dosage of active ingredient employed may vary depending on the particular compound employed, the mode of administration and the severity of the condition being treated. However, in general, satisfactory results are obtained when the compounds of the invention are administered at a daily dosage of from about 0.5 to about 500 mg/kg of animal body weight, preferably given in divided doses two to four times a day, or in a sustained release form. For most large mammals, the total daily dosage is from about 1 to 100 mg, preferably from about 2 to 80 mg. Dosage forms suitable for internal use comprise from about 0.5 to 500 mg of the active compound in

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intimate admixture with a solid or liquid pharmaceutically acceptable carrier. This dosage regimen may be adjusted to provide the optimal therapeutic response. For example, several divided doses may be administered daily or the dose may be proportionally reduced as indicated by the exigencies of the therapeutic situation.

5           These active compounds may be administered orally as well as by intravenous, intramuscular, or subcutaneous routes. Solid carriers include starch, lactose, dicalcium phosphate, microcrystalline cellulose, sucrose and kaolin, while liquid carriers include sterile water, polyethylene glycols, non-ionic surfactants and edible oils such as corn, peanut and sesame oils, as are appropriate to the nature of the active ingredient and the  
10           particular form of administration desired. Adjuvants customarily employed in the preparation of pharmaceutical compositions may be advantageously included, such as flavoring agents, coloring agents, preserving agents, and antioxidants, for example, vitamin E, ascorbic acid, BHT and BHA.

          The preferred pharmaceutical compositions from the standpoint of ease of  
15           preparation and administration are solid compositions, particularly tablets and hard-filled or liquid-filled capsules. Oral administration of the compounds is preferred.

          These active compounds may also be administered parenterally or intraperitoneally. Solutions or suspensions of these active compounds as a free base or pharmacologically acceptable salt can be prepared in water suitably mixed with a  
20           surfactant such as hydroxypropylcellulose. Dispersions can also be prepared in glycerol, liquid, polyethylene glycols and mixtures thereof in oils. Under ordinary conditions of storage and use, these preparations contain a preservative to prevent the growth of microorganisms.

          The pharmaceutical forms suitable for injectable use include sterile aqueous  
25           solutions or dispersions and sterile powders for the extemporaneous preparation of sterile injectable solutions or dispersions. In all cases, the form must be sterile and must be fluid to the extent that easy syringe ability exists. It must be stable under conditions of manufacture and storage and must be preserved against the

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contaminating action of microorganisms such as bacterial and fungi. The carrier can be a solvent or dispersion medium containing, for example, water, ethanol (e.g., glycerol, propylene glycol and liquid polyethylene glycol), suitable mixtures thereof, and vegetable oil.

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The following non-limiting examples illustrate preparation of compounds of the invention.

#### EXAMPLE 1

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##### 1-(4-Amino-3'-chloro-biphenyl-3-yl)-ethanone

A mixture of 2-amino-5-bromo-*N*-methoxy-*N*-methylbenzamide (7.78g, 30 mmol), 3-chlorophenyl boronic acid (5.63g, 36 mmol), tetrakis(triphenylphosphine)palladium (0) (1.73g, 1.5 mmol), and sodium carbonate (7.63g, 72 mmol) in a mixture of DME and water (150 mL/30 mL) was degassed to remove the oxygen and was then heated at 85 °C under nitrogen for 3 hours. The reaction mixture was cooled to room temperature and treated with brine (30 mL) and ethyl acetate (100 mL). The organic layer was separated and aqueous layer was extracted with ethyl acetate (3x40 mL). The combined organic layers were washed with brine and dried with MgSO<sub>4</sub>. After removal of solvent, the residue was purified by a flash chromatography (silica gel, hexane:ethyl acetate/1:1) to give 5-(3-chlorophenyl)-*N*-methoxy-*N*-methylbenzamide as a brown oil (5g, 57%). To a solution of this benzamide (5g, 17.2 mmol) in anhydrous THF was added in a dropwise fashion a solution of methyllithium in ether (1.4M, 28.6 mL, 40 mL) at -78 °C under nitrogen. After stirring for 30 minutes, the reaction mixture was treated with a saturated aqueous ammonium chloride solution (50 mL) at -78 °C. Ethyl acetate (100 mL) was added, the organic layer was separated, and the aqueous layer was extracted with ethyl acetate (3x20 mL). The combined organic layers were washed (brine) and dried (MgSO<sub>4</sub>). After removal of solvent, the residue was purified by a

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flash chromatography (silica gel, hexane:ethyl acetate/2:1) to afford 1-(4-amino-3'-chloro-biphenyl-3-yl)-ethanone as yellow solid (2g, 47%); mp 89-90 °C; <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ 7.89 (d, 1H, *J* = 2.0 Hz), 7.51 (m, 2H), 7.25-7.40 (m, 3H), 6.73 (d, 1H, *J* = 8.6 Hz), 6.38 (br, 2H), 2.65 (s, 3H); MS (EI) *m/z* 268([M+Na]<sup>+</sup>, 60%); Anal. Calc.

5 For C<sub>14</sub>H<sub>12</sub>ClNO: C, 68.44, H, 4.92, N, 5.70. Found: C, 68.40, H, 4.89, N, 5.61.

## EXAMPLE 2

### 1-(4-Amino-3'-chloro-biphenyl-3-yl)-dimethyl-methanol

To a solution of 1-(4-amino-3'-chloro-biphenyl-3-yl)-ethanone (.55g, 2.2  
10 mmol) in anhydrous THF under nitrogen was added a solution of methyl magnesium bromide (3.0 M in diethyl ether, 1 mL, 3 mmol) at 0 °C. The mixture was slowly warmed to room temperature and kept stirring under nitrogen for 18 hours. The mixture was treated with 10 mL of saturated ammonium chloride aqueous solution and ethyl acetate (50 mL) was added. The organic layer was separated and the aqueous  
15 layer was extracted with ethyl acetate (3x15 mL). The combined organic layers were washed with brine (20 mL) and dried (MgSO<sub>4</sub>). After removal of the solvent, the residue was purified by a flash chromatography (silica gel, hexane:ethyl acetate/2:1) to afford the title compound as an off-white solid: 186-188 °C (HCl salt). Anal. Calc. For C<sub>15</sub>H<sub>17</sub>Cl<sub>2</sub>NO: C, 60.42, H, 5.75, N, 4.7. Found: C, 60.51, H, 5.62, N, 4.56.

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## EXAMPLE 3

### 6-(3-Chloro-phenyl)-2,4,4-trimethyl-2-trifluoromethyl-1,4-dihydro-2H-benzo[d][1,3]oxazine

A mixture of (4-amino-3'-chloro-biphenyl-3-yl)-dimethyl-methanol (0.25 g,  
25 0.95 mmol), trifluoromethylacetone (0.16 g, 1.43 mmol), and *p*-toluenesulfonic acid (0.01 g, 0.05 mmol) in dry toluene (5 mL) was stirred under a blanket of nitrogen for 48 hours. Upon completion of the reaction, the toluene was removed and the residue purified via flash chromatography (silica gel, 10% ethyl acetate/hexane) to give 6-(3-

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chloro-phenyl)-2,4,4-trimethyl-2-trifluoromethyl-1,4-dihydro-2H-benzo[d][1,3]-oxazine (0.22 g, 65%) as a clear oil. The oil was dissolved in ether at -78 °C and then treated with a solution of 1N HCl in ether to give the hydrochloride salt of the title compound as a white solid: <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 7.67 (bs, 1H), 7.58 (d, 1H, *J* = 7.88 Hz), 7.42 (m, 3H), 7.32 (d, 1H, *J* = 7.96 Hz), 6.84 (d, 1H, *J* = 8.03 Hz), 1.52 (s, 3H), 1.5 (s, 6H); MS (APCI) *m/z* 354 ([M-H]<sup>+</sup>, 100%); Anal. Calc. For C<sub>18</sub>H<sub>17</sub>ClF<sub>3</sub>NO: C, 55.12; H, 4.62; N, 3.57. Found: C, 54.97; H, 4.59; N, 3.41.

#### EXAMPLE 4

##### 10 6-(3-Chloro-phenyl)-2,2,4,4-tetramethyl-1,4-dihydro-2H-benzo[d][1,3]oxazine

A mixture of (4-amino-3'-chloro-biphenyl-3-yl)-dimethyl-methanol (0.46 g, 1.8 mmol), acetone (0.16 g, 2.7 mmol), and *p*-toluenesulfonic acid (0.017 g, 0.09 mmol) in dry toluene (6 mL) was heated at 33 °C under a blanket of nitrogen overnight. Upon completion of the reaction, the toluene was removed and the compound purified via a  
15 flash chromatography (silica gel, 15% ethyl acetate/hexane) to give 6-(3-chloro-phenyl)-2,2,4,4-tetramethyl-1,4-dihydro-2H-benzo[d][1,3]oxazine (0.36 g, 68%) as a yellow oil. The oil was dissolved in ether at -78 °C and then treated with a solution of 1N HCl in ether to give the hydrochloride salt of the title compound as a yellow solid: <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 7.70 (bs, 1H), 7.61 (d, 1H, *J* = 7.82 Hz), 7.52 (bs, 1H), 7.44  
20 (m, 2H), 7.33 (d, 1H, *J* = 8.21 Hz), 6.87 (d, 1H, *J* = 7.85 Hz), 1.5 (s, 6H), 1.4 (s, 6H); MS (ESI) *m/z* 302 ([M+H]<sup>+</sup>, 100%); Anal. Calc. For C<sub>18</sub>H<sub>20</sub>ClNO: C, 63.91; H, 6.26; N, 4.14. Found: C, 64.08; H, 6.43; N, 4.14.

#### EXAMPLE 5

##### 25 6-(3-Nitro-phenyl)-2,2,4,4-trimethyl-1,4-dihydro-2H-benzo[d][1,3]oxazine

Prepared from 1-(4-amino-3'-nitro-biphenyl-3-yl)-ethanol and acetone in the same fashion as that of Example 4. Yellow solid: mp 188-189 °C; Anal. Calc. For C<sub>17</sub>H<sub>18</sub>N<sub>2</sub>O<sub>3</sub>·0.35 H<sub>2</sub>O: C, 67.02; H, 6.19; N, 9.20. Found: C, 66.7; H, 5.89; N, 9.03.

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**EXAMPLE 6****4-Amino-3'-chloro-biphenyl-3-carbonitrile**

A mixture of 2-amino-5-bromobenzonitrile (10g, 50 mmol), 3-chlorophenyl boronic acid (9.5g, 60 mmol), tetrakis(triphenylphosphine)-palladium (0) (3.5g, 3 mmol), and sodium carbonate (13g, 120 mmol) in a mixture of DME and water (100 mL/25 mL) was degassed to remove the oxygen and then heated to 85 °C under a blanket of nitrogen for 5 hours. The reaction mixture was cooled to ambient temperature and quenched with a saturated aqueous ammonium chloride solution (80 mL). Ethyl acetate (200 mL) was added, the organic layer was separated, and the aqueous layer was extracted with ethyl acetate (3x40 mL). The combined organic layers were washed with brine and dried with MgSO<sub>4</sub>. The solvent was removed *in vacuo* and the residue was purified by a silica gel flash chromatography (hexane:ethyl acetate/4:1) to afford 4-amino-3'-chloro-biphenyl-3-carbonitrile as an off-white solid (8g, 87%): mp 118-119 °C; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 7.80 (d, 1H, *J* = 2.3 Hz), 7.65-7.72 (m, 2H), 7.57 (d, 1H, *J* = 8.0 Hz), 7.42 (t, 1H, *J* = 7.9 Hz), 7.31 (m, 1H), 6.87 (d, 1H, *J* = 8.7 Hz), 6.29 (br, 2H); Anal. Calc. For C<sub>13</sub>H<sub>9</sub>ClN<sub>2</sub>: C, 68.28, H, 3.97, N, 12.25. Found: C, 67.68, H, 4.06, N, 11.89.

**EXAMPLE 7****6-(3-Chlorophenyl)-4-cyclopropyl-1H-quinazolin-2-one**

Prepared using a similar procedure as described in the literature (Tucker et al. *J. Med. Chem.*, **1994**, 37, 2437-2444). To a solution of cyclopropylmagnesium bromide prepared from magnesium (0.9g, 37 mmol) and cyclopropyl bromide (3.2 mL, 40 mmol) in anhydrous THF was added at 50 °C under nitrogen a solution of 4-amino-3'-chloro-biphenyl-3-carbonitrile (2.3g, 10 mmol) in anhydrous THF. After addition, the reaction mixture was kept at 50 °C for 30 minutes under nitrogen and treated with dimethyl carbonate in a dropwise manner. The reaction solution was stirred at 50 °C

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under nitrogen for 30 minutes and cooled to ambient temperature. A saturated aqueous ammonium chloride solution (30 mL) was added followed by addition of ethyl acetate (80 mL). The organic layer was separated and the aqueous layer was extracted with ethyl acetate (3x40 mL). The combined organic layers were washed with brine and dried with MgSO<sub>4</sub>. After removal of the solvent, the residue was purified by a flash chromatography (silica gel, methylene chloride:methanol/25:1) to give 6-(3-chlorophenyl)-4-cyclopropyl-1H-quinazolin-2-one as a yellowish solid (0.55g, 18%): mp 189-190 °C; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 11.71 (s, 1H, D<sub>2</sub>O exchangeable), 8.56 (d, 1H, *J* = 1.3 Hz), 8.09 (dd, 1H, *J* = 8.6, 1.6 Hz), 7.92 (s, 1H), 7.77 (d, 1H, *J* = 7.7 Hz), 7.52 (t, 1H, *J* = 7.9 Hz), 7.45 (d, 1H, *J* = 8.1 Hz), 7.36 (d, 1H, *J* = 8.6 Hz), 3.15 (m, 1H), 1.20 (m, 4H); MS (CI) *m/z* 297 ([M+H]<sup>+</sup>, 100%); Anal. Calc. For C<sub>17</sub>H<sub>13</sub>ClN<sub>2</sub>O: C, 69.98, H, 4.49, N, 9.23. Found: C, 67.98, H, 4.46, N, 9.10.

#### EXAMPLE 8

##### 15 6-(3-Chlorophenyl)-4-cyclopropyl-1-(4-methoxybenzyl)-1H-quinazolin-2-one

To a suspension of 6-(3-chlorophenyl)-4-cyclopropyl-1H-quinazolin-2-one (0.5g, 1.68 mmol) in anhydrous DMF was added potassium hexamethylsilyl amide (0.45g, 2.1 mmol) at ambient temperature under nitrogen. The reaction mixture was stirred at ambient temperature for 30 minutes, treated with *p*-methoxy benzyl chloride (0.35 mL, 2.5 mmol), and heated at 55 °C for 5 hours. The mixture was then cooled to room temperature and quenched with a saturated aqueous ammonium chloride solution (10 mL). Methylene chloride (50 mL) was added and organic layer was separated. The aqueous layer was extracted with methylene chloride (2x20 mL) and the combined organic layers were washed with brine and dried (MgSO<sub>4</sub>). After removal of solvent, the residue was separated on a flash chromatography (silica gel, hexane:ethyl acetate/1:1) to afford 6-(3-chlorophenyl)-4-cyclopropyl-1-(4-methoxybenzyl)-1H-quinazolin-2-one as an off-white solid: mp 173-174 °C; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 8.65 (d, 1H, *J* = 1.8 Hz), 8.10 (dd, 1H, *J* = 8.9, 1.8 Hz), 7.93 (d, 1H, *J*



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= 1.6 Hz), 7.79 (d, 1H,  $J = 7.6$  Hz), 7.53 (d, 2H,  $J = 8.4$  Hz), 7.46 (t, 1H,  $J = 8.1$  Hz), 7.21 (d, 2H,  $J = 8.6$  Hz), 6.88 (d, 2H,  $J = 8.6$  Hz), 5.41 (s, 2H), 3.73 (s, 3H), 3.18 (m, 1H), 1.18-1.27 (m, 4H); MS (CI)  $m/z$  417( $[M+H]^+$ , 100%); Anal. Calc. For  $C_{25}H_{21}ClN_2O_2$ : C, 72.02, H, 5.08, N, 6.72. Found: C, 71.88, H, 4.91, N, 6.70.

5        6-(3-Chlorophenyl)-4-cyclopropyl-2-(4-methoxybenzyloxy)quinazoline was obtained as a side product, off-white solid: mp 158-159 °C;  $^1H$ -NMR (DMSO- $d_6$ )  $\delta$  8.75 (d, 1H,  $J = 1.7$  Hz), 8.26 (dd, 1H,  $J = 8.8, 1.8$  Hz), 8.01 (s, 1H), 7.84 (m, 2H), 7.56 (t, 1H,  $J = 7.9$  Hz), 7.45 (m, 3H), 6.96 (d, 2H,  $J = 8.6$  Hz), 5.38 (s, 2H), 3.75 (s, 3H), 3.24 (m, 1H), 1.25 (m, 4H); MS (CI)  $m/z$  417( $[M+H]^+$ , 100%); Anal. Calc. For  $C_{25}H_{21}ClN_2O_2$ : C, 72.02, H, 5.08, N, 6.72. Found: C, 72.19, H, 4.91, N, 6.65.

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#### EXAMPLE 9

##### 6-(3-Chlorophenyl)-4-cyclopropyl-4-methyl-3,4-dihydro-1H-quinazolin-2-one

15        To a solution of 6-(3-chlorophenyl)-4-cyclopropyl-1-(4-methoxybenzyl)-1H-quinazolin-2-one (0.25g, 0.6 mmol) in anhydrous ether was added, at ambient temperature under nitrogen, magnesium triflate (0.78g, 2.4 mmol). The mixture was stirred for 30 minutes and treated with a solution of methylmagnesium bromide in ether (3.0 M, 1.0 mL, 3.0 mmol). The reaction mixture was kept at room temperature

20        under nitrogen for 3 hours and quenched with a mixture of a saturated aqueous ammonium chloride solution (10 mL) and 1N aqueous HCl solution (5 mL). The mixture was stirred at room temperature for 20 minutes and ethyl acetate (40 mL) was added. The organic layer was separated and the aqueous layer was extracted with ethyl acetate (3x10 mL). The combined organic layers were washed with brine and

25        dried ( $MgSO_4$ ). The solvent was removed and a half portion of the residue was dissolved in TFA (3 mL) and was stirred under nitrogen at room temperature for 72 hours. The solution was poured onto ice-water and the white precipitate obtained was collected on a filter. The solid was washed with water and then purified by a flash

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chromatography (methylene chloride:methanol/25:1, silica gel) to give 6-(3-chlorophenyl)-4-cyclopropyl-4-methyl-3,4-dihydro-1H-quinazolin-2-one as off-white solid (40 mg, 43%): mp 125-127 °C; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 9.21 (s, 1H), 7.71 (s, 1H), 7.61 (d, 1H, *J* = 7.8 Hz), 7.55 (d, 1H, *J* = 1.6 Hz), 7.47 (dd, 1H, *J* = 8.3, 1.8 Hz), 7.45 (t, 1H, *J* = 7.8 Hz), 7.36 (d, 1H, *J* = 8.1 Hz), 6.84 (d, 1H, *J* = 8.2 Hz), 6.79 (s, 1H), 1.54 (s, 3H), 1.11 (m, 1H), 0.42 (m, 1H), 0.15-0.20 (m, 3H); MS (ESI) *m/z* 313([M+H]<sup>+</sup>, 100%).

#### EXAMPLE 10

##### 10 6-(3-Chlorophenyl)-4-cyclopropyl-3,4-dimethyl-3,4-dihydro-1H-quinazolin-2-one

To a solution of half of the crude addition product from Example 9 in anhydrous DMF (5 mL) was added under nitrogen at room temperature sodium hydride (25 mg, 0.63 mmol). The mixture was stirred at ambient temperature for 30 minutes and treated with methyl iodide (0.5 mL, excess). After stirring for 3.5 hours, 15 a mixture of saturated aqueous ammonium chloride and 1N HCl aqueous solution (10 mL/5 mL) was added to the reaction mixture. Ethyl acetate (30 mL) was added and organic layer was separated. The aqueous layer was extracted with ethyl acetate (3x10 mL) and the combined organic layers were washed with brine and dried (MgSO<sub>4</sub>). The solvent was removed and residue was dissolved in a mixture of methylene chloride and 20 TFA (2 mL/2 mL). After stirring for 3 hours, the solution was poured onto ice-water and neutralized by addition of a saturated aqueous sodium bicarbonate solution. The ethyl acetate (30 mL) was added and organic layer was separated. The aqueous layer was extracted with ethyl acetate (3x10 mL). The combined organic layers were washed with brine and dried (MgSO<sub>4</sub>). The solvent was removed *in vacuo* and the 25 residue was purified by a flash chromatography (silica gel, hexane:ethyl acetate/1:1) to afford 6-(3-chlorophenyl)-4-cyclopropyl-3,4-dimethyl-3,4-dihydro-1H-quinazolin-2-one as a white solid (11 mg, 11% for three steps): mp 193-194 °C; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 9.51 (s, 1H), 7.68 (s, 1H), 7.59 (d, 1H, *J* = 8.0 Hz), 7.57 (d, 1H, *J* = 1.6 Hz),

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7.51 (dd, 1H,  $J = 8.3, 1.7$  Hz), 7.46 (t, 1H,  $J = 7.8$  Hz), 7.35 (d, 1H,  $J = 8.1$  Hz), 6.87 (d, 1H,  $J = 8.3$  Hz), 3.02 (s, 3H), 1.51 (s, 3H), 1.25 (m, 1H), 0.32-0.51 (m, 3H), 0.25 (m, 1H); MS (CI)  $m/z$  327 ( $[M+H]^+$ , 100%). Anal. Calc. For  $C_{19}H_{19}ClN_2O \cdot 0.3 H_2O$ : C, 68.69, H, 5.95, N, 8.43. Found: C, 68.69, H, 5.70, N, 8.18.

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**EXAMPLE 11****6-(3-Chloro-phenyl)-4,4-dimethyl-3,4-dihydro-1H-quinolin-2-one**

Sodium hydride (1.16 g, ca 29 mmol, 60% in oil) was washed with hexane (x3) then placed in anhydrous THF (50 mL) under nitrogen. To this slurry was then added dropwise a solution of 4-bromoaniline (5.0 g, 29 mmol) in dry THF (100 ml). After 10 0.5 h, the mixture was cooled to 0 °C and treated with a solution of 3,3'-dimethylacryloyl chloride in anhydrous THF (30 ml). After 4 h, the mixture was quenched with saturated aqueous ammonium chloride solution, and extracted with diethyl ether. The organic layer was washed with water, brine, dried ( $Na_2SO_4$ ) and 15 evaporated. The residue was then crystallized from EtOAc/hexane to afford *N*-(3,3'-dimethylacryloyl)-4-bromoaniline (3.36 g, 13.22 mmol, 46%) as a white solid:  $^1H$  NMR ( $CDCl_3$ )  $\delta$  7.42 (s, 4H), 7.06 (s, 1H), 5.68 (s, 1H), 2.21 (s, 3H), 1.09 (s, 3H); MS (EI)  $m/z$  253 [ $M^+$ ].

*N*-(3,3'-Dimethylacryloyl)-4-bromoaniline (4.0 g, 15.38 mmol) was heated 20 under nitrogen to ca. 130 - 140 °C causing the solid to melt. Aluminum chloride (3.07 g, 23 mmol) was added and heating continued. After 1 h, the mixture was cooled and quenched carefully with water and then extracted with dichloromethane (3x60 mL). The combined organic layers were washed with water, dried ( $MgSO_4$ ) and evaporated. The residue was then subjected to column chromatography ( $SiO_2$ , EtOAc/hexane 25 gradient elution) to afford 6-bromo-4,4-dimethyl-3,4-dihydro-1H-quinolin-2-one (1.69g, 6.6 mmol, 47%) as a white solid:  $^1H$ -NMR ( $CDCl_3$ )  $\delta$  8.82 (s, 1H), 7.39 (d, 1H,  $J = 2$  Hz), 7.29 (dd, 1H,  $J = 8.3, 2.0$  Hz), 6.71 (d, 1H,  $J = 8.0$  Hz), 2.47 (s, 2H), 1.32 (s, 6H).

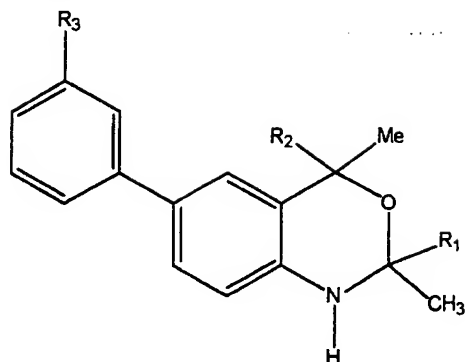
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To a solution of the last cited compound (0.5 g, 1.96 mmol) in dimethoxyethane (15 mL) was added tetrakis(triphenylphosphine)palladium (0) (0.11 g, 0.09 mmol) under nitrogen. After 15 min., 3-chlorophenylboronic acid (0.6 g, 3.9 mmol) was added followed by potassium carbonate (1.62 g, 11.7 mmol) in water (7.5 mL). After 1.5 h at reflux, the mixture was cooled, filtered, and extracted with EtOAc. The organic layer was washed with water, dried (MgSO<sub>4</sub>) and evaporated. The residue was then subjected to column chromatography (SiO<sub>2</sub>, EtOAc:hexane/3:1) and crystallized from dichloromethane/hexane to afford 6-(3-chloro-phenyl)-4,4-dimethyl-3,4-dihydro-1H-quinolin-2-one (0.21 g, 0.73 mmol, 37%) as a white solid; mp. 211 - 215 °C; <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ 8.48 (s, 1H), 7.53 (s, 1H), 7.47 (d, 1H, *J* = 2.0 Hz), 7.44 - 7.29 (m, 4H), 6.87 (1H, d, *J* = 2.0 Hz), 2.54 (s, 2H), 1.39 (s, 6H); MS ((-)ES) *m/z* 284 [M-H]<sup>+</sup>.

#### EXAMPLE 12 – Pharmacology

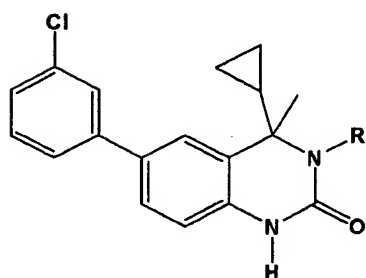
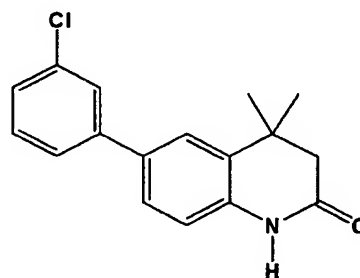
The compounds of this invention were tested in the relevant assay as described below and their potency are in the range of 0.01 nM to 5 μM in the *in vitro* assays and 0.001 to 300 mg/kg in the *in vivo* assays. The selected examples are listed below

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**Table 1**

Compound	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	hPR CV-1
				IC <sub>50</sub> (nM)
1	Me	Me	Cl	10.0
2	CF <sub>3</sub>	Me	Cl	50.0
3	Me	H	NO <sub>2</sub>	1675

5

R = H, hPR CV-1, IC<sub>50</sub> = 1075 nMR = Me, hPR CV-1, IC<sub>50</sub> = 580 nMhPR CV-1, IC<sub>50</sub> = 1075 nM

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#### A. In-vitro biology

The in-vitro biology is determined by (1) competitive Radioligand Binding: using the A-form of the human progesterone receptor with progesterone as the radioligand; (2) co-transfection assay, which provides functional  
5 activity expressed as agonist EC50 and Antagonist IC50 values; (3) a T47D cell proliferation, which is a further functional assay which also provides agonist and antagonist data; and (4) T47D cell alkaline phosphatase assay, which is a further functional assay which also provides agonist and antagonist data.

1. hPR Binding assay - This assay is carried out in  
10 accordance with: Pathirana, C.; Stein, R.B.; Berger, T.S.; Fenical, W.; Ianiro, T.; Mais, D.E.; Torres, A.; Glodman, M.E., *Nonsteroidal human progesterone receptor modulators from the marine alga cymoplia barbata*, J. Steroid Biochem. Mol. Biol., 1992, 41, 733-738.

#### 2. PRE-luciferase assay in CV-1 cells

15 The object of this assay is to determine a compound's progestational or antiproggestational potency based on its effect on PRE-luciferase reporter activity in CV-1 cells co-transfected with human PR and PRE-luciferase plasmids. The materials methods used in the assay are as follows.

a. Medium: The growth medium was as follows:  
20 DMEM (BioWhittaker) containing 10% (v/v) fetal bovine serum (heat inactivated), 0.1 mM MEM non-essential amino acids, 100U/ml penicillin, 100mg/ml streptomycin, and 2 mM GlutaMax (GIBCO, BRL). The experimental medium was as follows:  
DMEM (BioWhittaker), phenol red-free, containing 10% (v/v) charcoal-stripped fetal bovine serum (heat-inactivated), 0.1 mM MEM non-essential amino acids, 100U/ml  
25 penicillin, 100mg/ml streptomycin, and 2 mM GlutaMax (GIBCO, BRL).

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b. Cell culture, transfection, treatment, and luciferase assay

Stock CV-1 cells are maintained in growth medium. Co-transfection is done using  $1.2 \times 10^7$  cells, 5 mg pLEM plasmid with hPR-B inserted at SphI and BamHI sites, 10 mg pGL3 plasmid with two PREs upstream of the luciferase sequence, and 50 mg sonicated calf thymus DNA as carrier DNA in 250 ml. Electroporation is carried out at 260 V and 1,000 mF in a Biorad Gene Pulser II. After electroporation, cells are resuspended in growth medium and plated in 96-well plate at 40,000 cells/well in 200  $\mu$ l. Following overnight incubation, the medium is changed to experimental medium. Cells are then treated with reference or test compounds in experimental medium. Compounds are tested for antiprogesterone activity in the presence of 3 nM progesterone. Twenty-four hr. after treatment, the medium is discarded, cells are washed three times with D-PBS (GIBCO, BRL). Fifty  $\mu$ l of cell lysis buffer (Promega, Madison, WI) is added to each well and the plates are shaken for 15 min in a Titer Plate Shaker (Lab Line Instrument, Inc.). Luciferase activity is measured using luciferase reagents from Promega.

c. Analysis of Results:

Each treatment consists of at least 4 replicates. Log transformed data are used for analysis of variance and nonlinear dose response curve fitting for both agonist and antagonist modes. Huber weighting is used to downweight the effects of outliers.  $EC_{50}$  or  $IC_{50}$  values are calculated from the retransformed values. JMP software (SAS Institute, Inc.) is used for both one-way analysis of variance and non-linear response analyses.

d. Reference Compounds:

Progesterone and trimegestone are reference progestins and RU486 is the reference antiprogesterone. All reference compounds are run in full dose-response curves and the  $EC_{50}$  or  $IC_{50}$  values are calculated.

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**Table 2. Estimated EC<sub>50</sub>, standard error (SE), and 95% confidence intervals (CI) for reference progestins from three individual studies**

5	Compound	Exp.	EC <sub>50</sub>	SE	95% CI	
			(nM)		lower	upper
	Progesterone	1	0.616	0.026	0.509	0.746
		2	0.402	0.019	0.323	0.501
		3	0.486	0.028	0.371	0.637
10	Trimegestone	1	0.0075	0.0002	0.0066	0.0085
		2	0.0081	0.0003	0.0070	0.0094
		3	0.0067	0.0003	0.0055	0.0082

**Table 3. Estimated IC<sub>50</sub>, standard error (SE), and 95% confident interval (CI) for the antiprogestin, RU486 from three individual studies**

15	Compound	Exp.	IC <sub>50</sub>	SE	95% CI	
			(nM)		lower	upper
20	RU486	1	0.028	0.002	0.019	0.042
		2	0.037	0.002	0.029	0.048
		3	0.019	0.001	0.013	0.027

Progestational activity: Compounds that increase PRE-luciferase activity significantly ( $p < 0.05$ ) compared to vehicle control are considered active.

25 Antiprogestational activity: Compounds that decrease 3 nM progesterone induced PRE-luciferase activity significantly ( $p < 0.05$ )

EC<sub>50</sub>: Concentration of a compound that gives half-maximal increase PRE-luciferase activity (default-nM) with SE.

30 IC<sub>50</sub>: Concentration of a compound that gives half-maximal decrease in 3 nM progesterone induced PRE-luciferase activity (default-nM) with SE.



a. Growth medium: DMEM:F12 (1:1)

**b. Treatment medium:** Minimum Essential Medium (MEM) (#51200-038 GIBCO, BRL) phenol red-free supplemented with 0.5% charcoal stripped fetal bovine serum, 100U/ml penicillin, 200 mg/ml streptomycin, and 2 mM GlutaMax (GIBCO, BRL).

### c. Cell culture

15 Stock T47 D cells are maintained in growth medium. For BrdU incorporation assay, cells are plated in 96-well plates (Falcon, Becton Dickinson Labware) at 10,000 cells/well in growth medium. After overnight incubation, the medium is changed to treatment medium and cells are cultured for an additional 24 hr before treatment. Stock compounds are dissolved in appropriate vehicle (100% ethanol or 50% ethanol/50% DMSO), subsequently diluted in treatment medium and added to the cells. Progestin and antiprogesterone reference compounds are run in full dose-response curves. The final concentration of vehicle is 0.1%. In control wells, cells receive vehicle only. Antiprogesterones are tested in the presence of 0.03 nM trimegestone, the reference progestin agonist. Twenty-four hours after treatment, the medium is discarded and cells are labeled with 10 mM BrdU (Amersham Life Science, Arlington Heights, IL) in treatment medium for 4 hr.

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d. Cell Proliferation Assay

At the end of BrdU labeling, the medium is removed and BrdU incorporation is measured using a cell proliferation ELISA kit (#RPN 250, Amersham Life Science) according to manufacturer's instructions. Briefly, cells are fixed in an ethanol containing fixative for 30 min, followed by incubation in a blocking buffer for 30 min to reduce background. Peroxidase-labeled anti-BrdU antibody is added to the wells and incubated for 60 min. The cells are rinsed three times with PBS and incubated with 3,3',5,5'-tetramethylbenzidine (TMB) substrate for 10-20 min depending upon the potency of tested compounds. Then 25  $\mu$ l of 1 M sulfuric acid is added to each well to stop color reaction and optical density is read in a plate reader at 450 nm within 5 min.

e. Analysis of Results:

Square root-transformed data are used for analysis of variance and nonlinear dose response curve fitting for both agonist and antagonist modes. Huber weighting is used to downweight the effects of outliers.  $EC_{50}$  or  $IC_{50}$  values are calculated from the retransformed values. JMP software (SAS Institute, Inc.) is used for both one-way analysis of variance and non-linear dose response analyses in both single dose and dose response studies.

f. Reference Compounds:

Trimegestone and medroxyprogesterone acetate (MPA) are reference progestins and RU486 is the reference antiprogesterin. All reference compounds are run in full dose-response curves and the  $EC_{50}$  or  $IC_{50}$  values are calculated.

**Table 4. Estimated EC<sub>50</sub>, standard error (SE), and 95% confidence intervals (CI) for individual studies**

5	<u>Compound</u>	<u>Exp</u>	<u>EC<sub>50</sub></u> <u>(nM)</u>	<u>SE</u>	<u>95% CI</u>	
					<u>lower</u>	<u>upper</u>
	Trimegestone	1	0.017	0.003	0.007	0.040
		2	0.014	0.001	0.011	0.017
		3	0.019	0.001	0.016	0.024
10	MPA	1	0.019	0.001	0.013	0.027
		2	0.017	0.001	0.011	0.024

**Table 5. Estimated IC<sub>50</sub>, standard error, and 95% confident interval for the antiprogesterin, RU486**

15	<u>Compound</u>	<u>Exp</u>	<u>IC<sub>50</sub></u> <u>(nM)</u>	<u>SE</u>	<u>95% CI</u>	
					<u>lower</u>	<u>upper</u>
20	RU486	1	0.011	0.001	0.008	0.014
		2	0.016	0.001	0.014	0.020
		3	0.018	0.001	0.014	0.022

EC<sub>50</sub>: Concentration of a compound that gives half-maximal increase in BrdU incorporation with SE; IC<sub>50</sub>: Concentration of a compound that gives half-maximal decrease in 0.1 trimegestone induced BrdU incorporation with SE

25

#### 4. T47D cell alkaline phosphatase assay

The purpose of this assay is to identify progestins or antiprogesterins by determining a compound's effect on alkaline phosphatase activity in T47D cells. The materials and methods used in this assay are as follows.

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a. Culture medium: DMEM:F12 (1:1) (GIBCO, BRL)

supplemented with 5% (v/v) charcoal stripped fetal bovine serum (not heat-inactivated), 100U/ml penicillin, 100 µg/ml streptomycin, and 2 mM GlutaMax (GIBCO, BRL).

b. Alkaline phosphatase assay buffer:

- 5 I. 0.1 M Tris-HCl, pH 9.8, containing 0.2% Triton X-100  
II. 0.1 M Tris-HCl, pH 9.8 containing 4 mM p-nitrophenyl phosphate (Sigma).

c. Cell Culture and Treatment:

Frozen T47D cells were thawed in a 37°C water bath and diluted to 280,000 cells/ml in culture medium. To each well in a 96-well plate (Falcon, Becton Dickinson Labware), 180 µl of diluted cell suspension was added. Twenty µl of reference or test compounds diluted in the culture medium was then added to each well. When testing for progestin antagonist activity, reference antiprogesterins or test compounds were added in the presence of 1 nM progesterone. The cells were  
15 incubated at 37°C in a 5% CO<sub>2</sub>/humidified atmosphere for 24 hr.

d. Alkaline Phosphatase Enzyme Assay:

At the end of treatment, the medium was removed from the plate and fifty µl of assay buffer I was added to each well. The plates were shaken in a titer plate shaker for 15 min. Then 150 µl of assay buffer II was added to  
20 each well. Optical density measurements were taken at 5 min intervals for 30 min at a test wavelength of 405 nM.

e. Analysis of Results: Analysis of dose-response data

For reference and test compounds, a dose response curve is generated for dose (X-axis) vs. the rate of enzyme reaction (slope) (Y-axis). Square  
25 root-transformed data are used for analysis of variance and nonlinear dose response curve fitting for both agonist and antagonist modes. Huber weighting is used to downweight the effects of outliers. EC<sub>50</sub> or IC<sub>50</sub> values are calculated from the retransformed values. JMP software (SAS Institute, Inc.) is used for both one-way

analysis of variance and non-linear dose response analyses in both single dose and dose response studies.

f. Reference Compounds:

Progesterone and trimegestone are reference progestins and RU486 is the  
5 reference antiprogesterin. All reference compounds are run in full dose response curves and the EC<sub>50</sub> or IC<sub>50</sub> values are calculated.

**Table 6. Estimated EC<sub>50</sub>, standard error (SE), and 95% confidence intervals (CI) for reference progestins from three independent experiments**

10

Compound	Exp.	EC <sub>50</sub> (nM)	SE	95% CI	
				lower	upper
Progesterone	1	0.839	0.030	0.706	0.996
	2	0.639	0.006	0.611	0.669
	3	1.286	0.029	1.158	1.429
Trimegestone	1	0.084	0.002	0.076	0.091
	2	0.076	0.001	0.072	0.080
	3	0.160	0.004	0.141	0.181

**Table 7. Estimated IC<sub>50</sub>, standard error, and 95% confident interval for the reference antiprogesterin RU486 from three independent experiments**

25

Compound	Exp	IC 50 (nM)	SE	95% CI	
				lower	upper
RU486	1	0.103	0.002	0.092	0.115
	2	0.120	0.001	0.115	0.126
	3	0.094	0.007	0.066	0.134

30

## B. In-vivo Biology

The primary in-vivo assay is the rat decidualization model which may be used to determine progestational effects of both agonists and antagonists. The secondary in-vivo assay is the rat ovulation inhibition model which is under development and hence the protocol is un-available.

1. Rat decidualization assay The objective of this procedure is used to evaluate the effect of progestins and antiprogestins on rat uterine decidualization and compare the relative potencies of various test compounds. The materials and methods used in this assay are as follows.

a. Methods: Test compounds are dissolved in 100% ethanol and mixed with corn oil (vehicle). Stock solutions of the test compounds in oil (Mazola<sup>TM</sup>) are then prepared by heating (~80 °C) the mixture to evaporate ethanol. Test compounds are subsequently diluted with 100% corn oil or 10% ethanol in corn oil prior to the treatment of animals. No difference in decidual response was found when these two vehicles were compared.

### b. Animals (RACUC protocol #5002)

Ovariectomized mature female Sprague-Dawley rats (~60-day old and 230g) are obtained from Taconic (Taconic Farms, NY) following surgery. Ovariectomy is performed at least 10 days prior to treatment to reduce circulating sex steroids. Animals are housed under 12 hr light/dark cycle and given standard rat chow and water ad libitum.

### c. Treatment

Rats are weighed and randomly assigned to groups of 4 or 5 before treatment. Test compounds in 0.2 ml vehicle are administered by subcutaneous injection in the nape of the neck or by lavage using 0.5 ml. The animals are treated once daily for seven days. For testing antiprogestins, animals are given the test compounds and a EC<sub>50</sub> dose of progesterone (5.6 mg/kg) during the first three days

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of treatment. Following decidual stimulation, animals continue to receive progesterone until necropsy four days later.

d. Dosing

Doses are prepared based upon mg/kg mean group body weight. In all studies, a control group receiving vehicle is included. Determination of dose-response curves is carried out using doses with half log increases (e.g. 0.1, 0.3, 1.0, 3.0 mg/kg).

e. Decidual induction

Approximately 24 hr after the third injection, decidualization is induced in one of the uterine horns by scratching the antimesometrial luminal epithelium with a blunt 21 G needle. The contralateral horn is not scratched and serves as an unstimulated control. Approximately 24 hr following the final treatment, rats are sacrificed by CO<sub>2</sub> asphyxiation and body weight measured. Uteri are removed and trimmed of fat. Decidualized (D-horn) and control (C-horn) uterine horns are weighed separately.

f. Analysis of Results:

The increase in weight of the decidualized uterine horn is calculated by D-horn/C-horn and logarithmic transformation is used to maximize normality and homogeneity of variance. The Huber M-estimator is used to down weight the outlying transformed observations for both dose-response curve fitting and one-way analysis of variance. JMP software (SAS Institute, Inc.) is used for both one-way ANOVA and non-linear dose-response analyses.

g. Reference Compounds:

All progestin reference compounds were run in full dose-response curves and the EC<sub>50</sub> for uterine wet weight were calculated.

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**Table 8. Estimated EC<sub>50</sub>, standard error (SE), and 95% confidence intervals for individual studies**

5	Compound	Exp	EC <sub>50</sub>		95% CI	
			(mg/kg, s.c.)	SE	lower	upper
	Progesterone	1	5.50	0.77	4.21	7.20
		2	6.21	1.12	4.41	8.76
	3-Ketodesogestrel	1	0.11	0.02	0.07	0.16
10		2	0.10	0.05	0.11	0.25
		3	0.06	0.03	0.03	0.14
15	Levonorgestrel	1	0.08	0.03	0.04	0.16
		2	0.12	0.02	0.09	0.17
		3	0.09	0.02	0.06	0.13
		4	0.09	0.02	0.06	0.14
20	MPA	1	0.42	0.03	0.29	0.60
		2	0.39	0.05	0.22	0.67
		3	0.39	0.04	0.25	0.61

**Table 9. Estimated average EC<sub>50</sub>, standard error, and 95% confidence intervals for dose-response curves of 3 reference compounds**

25

	Compound	EC <sub>50</sub>		95% CI	
		(mg/kg, s.c.)	SE	lower	upper
	Progesterone	5.62	0.62	4.55	7.00
	3-Ketodesogestrel	0.10	0.02	0.07	0.14
30	Levonorgestrel	0.10	0.01	0.08	0.12



**Table 10. Estimated IC<sub>50</sub>, standard error, and 95% confident interval for the antiprogestin, RU 486**

5	Compound	Exp.	IC <sub>50</sub> (mg/kg, p.o.)	SE	95% CI	
					lower	upper
	RU 486	1	0.21	0.07	0.05	0.96
		2	0.14	0.02	0.08	0.27

Concentration: Compound concentration in assay(default-mg/kg body weight)

10 Route of administration: Route the compound is administered to the animals

Body weight: Mean total animal body weight (default-kg)

D-horn: Wet weight of decidualized uterine horn (default-mg)

C-horn: Wet weight of control uterine horn (default-mg)

Decidual response: [(D-C)/C]x100%

15 Progestational activity: Compounds that induce decidualization significantly (p<0.05) compared to vehicle control are considered active

Antiprogesterational activity: Compounds that decrease EC<sub>50</sub> progesterone induced decidualization significantly (p<0.05)

20 EC<sub>50</sub> for uterine weight: Concentration of compound that gives half-maximal increase in decidual response (default-mg/kg)

IC<sub>50</sub> for uterine weight: Concentration of compound that gives half-maximal decrease in EC<sub>50</sub> progesterone induced decidual response (default-mg/kg)

25

### EXAMPLE 13

#### 5-(2,4,4-Trimethyl-1,4-dihydro-2H-3,1-benzoxazin-6-yl)thiophene-2-carbonitrile

A solution of 2-amino-5-bromobenzoic acid (10g, 46 mmol) in dry THF (200 mL) was treated at -78 °C under nitrogen with a solution of methylmagnesium bromide in ether (3.0 M, 90 mL, 270 mmol). The reaction mixture was slowly

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warmed to ambient temperature, kept stirring for 48 hours under nitrogen and then poured into a cold 0.5 N aqueous hydrochloride solution (300 mL). The mixture was neutralized with aqueous 1 N sodium hydroxide solution and ethyl acetate (300 mL) was added. The organic layer was separated and aqueous layer was extracted with ethyl acetate (3x100 mL). The combined organic layers were washed with brine and dried (MgSO<sub>4</sub>). After removal of solvent *in vacuo*, the residue was purified by a silica gel flash column chromatography(hexane:ethyl acetate/3:2) to give 2-(2-amino-5-bromophenyl)propan-2-ol as off-white solid (6g, 57%): mp 62-63 °C; <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ 7.19 (d, 1H, *J* = 2.3 Hz), 7.12 (dd, 1H, *J* = 8.4, 2.3 Hz), 6.51 (d, 1H, *J* = 8.4 Hz), 4.70 (s, 2H), 1.82 (s, 1H), 1.65 (s, 6H).

To a solution of 2-(2-amino-5-bromophenyl)propan-2-ol (27g, 125 mmol) in anhydrous toluene was added at ambient temperature under a blanket of nitrogen acetaldehyde (10.5 mL, 187 mmol). After 10 minutes, the mixture was passed through a pad of silica gel and filtrate was concentrated to yield 6-bromo-2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazine as off-white solid (25g, 78%): <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 7.22 (d, 1H, *J* = 2.2 Hz), 7.08 (dd, 1H, *J* = 8.6, 2.3 Hz), 6.51 (d, 1H, *J* = 8.6 Hz), 6.36 (s, 1H), 4.72 (m, 1H), 1.45 (s, 3H), 1.40 (s, 3H), 1.25 (d, 3H, *J* = 5.5 Hz).

A mixture of 6-bromo-2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazine (3.6g, 14 mmol), bis(pinacolato)diboron (5 g, 19.7 mmol), potassium acetate (4g, 41 mmol), and [1,1'-bis(diphenylphosphino)ferrocene]palladium (II) chloride (1:1 complex with methylene chloride, 0.4g, 0.5 mmol) in DMF (80 mL) was subject to a positive nitrogen flow to remove oxygen and then heated at 85 °C under a blanket of nitrogen for 18 hours. The mixture was allowed to cool to ambient temperature, treated with 5-bromo-2-thiophenecarbonitrile (4g, 21 mmol), [1,1'-bis(diphenylphosphino)ferrocene]palladium (II) chloride (1:1 complex with methylene chloride, 0.4g, 0.5 mmol), and aqueous sodium carbonate solution (2M, 35 mL, 70 mmol), and then heated at 85 °C under nitrogen for 3 hours. The reaction mixture was

- 57 -

allowed to cool to ambient temperature, brine (100 mL) and ethyl acetate (150 mL) were added. The organic layer was separated and aqueous layer was extracted with ethyl acetate (3x50 mL). The combined organic layers were dried (MgSO<sub>4</sub>) and concentrated. The residue was purified by a flash silica gel column

5 chromatography(THF:hexane/1:4) to afford the title compound as an off-white solid (1g, 25%): mp 172-173 °C; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 7.88 (d, 1H, *J* = 4.0 Hz), 7.47 (d, 1H, *J* = 4.0 Hz), 7.43 (d, 1H, *J* = 2.0 Hz), 7.32 (dd, 1H, *J* = 8.36, 2.4 Hz), 6.77 (s, 1H), 6.60 (d, 1H, *J* = 8.4 Hz), 4.83 (m, 1H), 1.51 (s, 3H), 1.48 (s, 3H), 1.28 (d, 3H, *J* = 5.6 Hz); MS (ESI) *m/z* 283 [M-H]<sup>+</sup>.

10

#### EXAMPLE 14

##### 3-Fluoro-5-(2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazin-6-yl)benzonitrile

Prepared according to the procedure for Example 13 from 6-bromo-2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazine and 3-bromo-5-fluorobenzonitrile. A white  
15 solid: mp 163-164 °C; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 8.02 (t, 1H, *J* = 1.5 Hz), 7.87 (dt, 1H, *J* = 10.6, 2.2 Hz), 7.65 (m, 1H), 7.55 (d, 1H, *J* = 2.2 Hz), 7.44 (dd, 1H, *J* = 8.4, 2.2 Hz), 6.63 (d, 1H, *J* = 8.4 Hz), 6.58 (s, 1H), 4.82 (m, 1H), 1.52 (s, 3H), 1.50 (s, 3H), 1.28 (d, 3H, *J* = 5.1 Hz); MS (ESI) *m/z* 295 [M-H]<sup>+</sup>.

20

#### EXAMPLE 15

##### 4-(2,4,4-Trimethyl-1,4-dihydro-2H-3,1-benzoxazin-6-yl)thiophene-2-carbonitrile

Prepared according to the procedure for Example 13 from 6-bromo-2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazine and 4-bromo-2-thiophenecarbonitrile. An  
off-white solid: mp 175-176 °C; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 8.39 (d, 1H, *J* = 1.5 Hz), 8.13  
25 (d, 1H, *J* = 1.5 Hz), 7.47 (d, 1H, *J* = 1.9 Hz), 7.36 (dd, 1H, *J* = 8.4, 1.9 Hz), 6.59 (d, 1H, *J* = 8.4 Hz), 6.41 (s, 1H), 4.78 (m, 1H), 1.51 (s, 3H), 1.47 (s, 3H), 1.28 (d, 3H, *J* = 5.4 Hz); MS (ESI) *m/z* 285 [M+H]<sup>+</sup>.

**EXAMPLE 16****4-Methyl-5-(2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazin-6-yl)thiophene-2-carbonitrile**

Prepared according to the procedure for Example 13 from 6-bromo-2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazine and 5-bromo-4-methyl-2-thiophenecarbonitrile. A yellowish solid: mp 145-146 °C; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 7.79 (s, 1H), 7.18 (d, 1H, *J* = 2.0 Hz), 7.13 (dd, 1H, *J* = 8.4, 2.0 Hz), 6.68 (s, 1H), 6.54 (d, 1H, *J* = 8.3 Hz), 4.83 (m, 1H), 2.26 (s, 3H), 1.49 (s, 3H), 1.46 (s, 3H), 1.28 (d, 3H, *J* = 5.5 Hz); MS (ESI) *m/z* 299 [M+H]<sup>+</sup>.

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**EXAMPLE 17****3-[(2R,4S)-2,4-Dimethyl-4-phenyl-1,4-dihydro-2H-3,1-benzoxazin-6-yl]-5-fluorobenzonitrile**

To a solution of 2-amino-4-bromobenzonitrile (5g, 25 mmol) in anhydrous THF, was added at 0 °C under nitrogen, phenylmagnesium bromide (3 M in ether, 25 mL, 75 mmol). The mixture was allowed to warm to room temperature, stirred under nitrogen for 15 hours, and treated with 2N aqueous hydrogen chloride solution (100 mL). The aqueous solution was heated to 50 °C for 3 hours, cooled to room temperature, and neutralized with a cold saturated sodium bicarbonate solution. Ethyl acetate (100 mL) was added and the organic layer was separated and aqueous layer was extracted with ethyl acetate (3x40 mL). The combined organic layers were dried (MgSO<sub>4</sub>) and evaporated. The residue was purified by a flash column chromatography (silica gel, hexane:ethyl acetate/4:1) to yield (2-amino-5-bromophenyl)(phenyl)methanone as a yellow crystal (2.13g, 31%): MS (ESI) *m/z* 276/278 [M+H]<sup>+</sup>.

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To a solution of (2-amino-5-bromophenyl)(phenyl)methanone (1g, 3.6 mmol) in anhydrous THF (15 mL) was added at room temperature under nitrogen methylmagnesium bromide (3M in ether, 3 mL, 9 mmol). After 3 hours, the mixture

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was treated with a saturated aqueous ammonium sulfate solution (30 mL) and ethyl acetate (50 mL). The organic layer was separated, dried (MgSO<sub>4</sub>), and evaporated. The residue was then dissolved in anhydrous toluene and treated at ambient temperature under nitrogen with acetaldehyde (2 mL). After 2 minutes, the solvent  
5 was removed and residue was purified by a column chromatography (silica gel, hexane:ethyl acetate/4:1) to afford 6-bromo-2,4-dimethyl-4-phenyl-1,4-dihydro-2H-3,1-benzoxazine as a yellowish solid (0.8g, 70%): MS (ESI) *m/z* 318/320 [M+H]<sup>+</sup>.

A mixture of 6-bromo-2,4-dimethyl-4-phenyl-1,4-dihydro-2H-3,1-benzoxazine (0.6g, 1.9 mmol), 3-cyano-5-fluorobenzene boronic acid (0.45g, 2.7 mmol),  
10 tetrakis(triphenylphosphine)palladium (0) (0.2g, 0.17 mmol), sodium carbonate (0.6g, 5.7 mmol) in a mixture of DME and water (20/5 mL) was subject to a positive nitrogen flow to remove oxygen and then heated to 85 °C under a blanket of nitrogen for 2 hours. The mixture was allowed to cool to ambient temperature. Brine (30 mL) and ethyl acetate (100 mL) were added. The organic layer was separated and aqueous  
15 layer was extracted with ethyl acetate (3x30 mL). The combined organic layers were dried (MgSO<sub>4</sub>), and evaporated. The residue was purified by a silica gel column chromatography (hexane:ethyl acetate/4:1) to afford the title compound as off-white solid (0.09g, 13%): mp 128-129 °C; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 8.08 (s, 1H), 7.91 (dt, 1H, *J* = 10.7, 1.9 Hz), 7.71 (d, 1H, *J* = 2.0 Hz), 7.65 (m, 1H), 7.57 (dd, 1H, *J* = 8.8,  
20 2.4 Hz), 7.32-7.36 (m, 2H), 7.24-7.29 (m, 3H), 6.71 (d, 1H, *J* = 1.6 Hz), 6.69 (d, 1H, *J* = 8.3 Hz), 4.33 (m, 1H), 1.84 (s, 3H), 1.24 (d, 3H, *J* = 5.5 Hz); MS (ESI) *m/z* 357 [M-H]<sup>-</sup>.

#### EXAMPLE 18

25 *tert*-Butyl 2-cyano-5-(2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazin-6-yl)-1H-pyrrole-1-carboxylate

*tert*-Butyl-5-(2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazin-6-yl)-1H-pyrrole-1-carboxylate was prepared according to the coupling procedure for Example 17 from

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6-bromo-2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazine and 1-*t*-butoxycarbonylpyrrol-2-yl boronic acid. To a solution of *tert*-butyl-5-(2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazin-6-yl)-1H-pyrrole-1-carboxylate (1g, 2.9 mmol) in anhydrous THF (20 mL) was added at -78 °C under nitrogen chlorosulfonyl isocyanate (0.35 mL, 4.0 mmol). The mixture was kept stirring at -78 °C under nitrogen for 2 hours, treated with anhydrous DMF (5 mL), and allowed to warm to room temperature. Aqueous ammonium sulfate solution (50 mL) and ethyl acetate (100 mL) was added and organic layer was separated, dried (MgSO<sub>4</sub>), and evaporated. The residue was purified by a silica gel column chromatography(hexane:ethyl acetate/4:1) to afford the title compound as a white solid (0.019g, 1.78%): <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 7.48 (d, 1H, *J* = 2.0 Hz), 7.36 (dd, 1H, *J* = 8.3, 2.0 Hz), 7.32 (d, 1H, *J* = 3.6 Hz), 7.14 (d, 1H, *J* = 8.3 Hz), 6.46 (d, 1H, *J* = 4.0 Hz), 5.35 (q, 1H, *J* = 5.2 Hz), 1.58 (d, 3H, *J* = 5.6 Hz), 1.56 (s, 3H), 1.51 (s, 3H), 1.38 (s, 9H); MS (ESI) *m/z* 366 [M-H].

15

#### EXAMPLE 19

**9H-Fluoren-9-ylmethyl-6-[1-(*tert*-butoxycarbonyl)-5-cyano-1H-pyrrol-2-yl]-2,4,4-trimethyl-2H-3,1-benzoxazine-1(4H)-carboxylate**

A mixture of *tert*-butyl-5-(2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazin-6-yl)-1H-pyrrole-1-carboxylate (1.7g, 4.96 mmol), 9-fluorenylmethyl chloroformate (1.92g, 7.5 mL), sodium carbonate (4g, 37 mmol) in dioxane (50 mL) and water (50 mL) was stirred at room temperature under a blanket of nitrogen for 6 hours. Ethyl acetate (100 mL) was added and organic layer was separated, dried (MgSO<sub>4</sub>), and evaporated. The residue was purified by a silica gel column chromatography(hexane:ethyl acetate/6:1) to afford 9H-fluoren-9-ylmethyl-6-[1-(*tert*-butoxycarbonyl)-1H-pyrrol-2-yl]-2,4,4-trimethyl-2H-3,1-benzoxazine-1(4H)-carboxylate as a clean oil.

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Using the procedure for Example 18, the title compound (0.8g, 65%) was prepared from 9H-fluoren-9-ylmethyl-6-[1-(tert-butoxycarbonyl)-1H-pyrrol-2-yl]-2,4,4-trimethyl-2H-3,1-benzoxazine-1(4H)-carboxylate (1.2g, 2.1 mmol) and chlorosulfonyl isocyanate (0.28 mL, 3.1 mmol). A white solid: mp 135-136 °C; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 7.90 (t, 2H, *J* = 6.7 Hz), 7.64 (d, 1H, *J* = 7.5 Hz), 7.59 (d, 1H, *J* = 7.1 Hz), 7.40 (td, 2H, *J* = 7.2, 2.0 Hz), 7.31-7.34 (m, 3H), 7.29 (d, 1H, *J* = 1.2 Hz), 7.04-7.09 (m, 2H), 6.44 (d, 1H, *J* = 3.57 Hz), 5.30 (q, 1H, *J* = 5.6 Hz), 4.86 (dd, 1H, *J* = 10.7, 5.2 Hz), 4.64 (dd, 1H, *J* = 10.8, 5.2 Hz), 4.33 (t, 1H, *J* = 4.7 Hz), 1.50 (s, 3H), 1.30 (s, 9H), 1.20 (s, 3H), 1.03 (d, 3H, *J* = 5.6 Hz); MS (ESI) *m/z* 590 [M+H]<sup>+</sup>.

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**EXAMPLE 20****5-(2,4,4-Trimethyl-1,4-dihydro-2H-3,1-benzoxazin-6-yl)-1H-pyrrole-2-carbonitrile**

9H-Fluoren-9-ylmethyl-6-[1-(tert-butoxycarbonyl)-5-cyano-1H-pyrrol-2-yl]-2,4,4-trimethyl-2H-3,1-benzoxazine-1(4H)-carboxylate (0.5g, 0.84 mmol) was heated under a blanket of nitrogen at 160 °C until gas evolution ceased. After cooling to room temperature, 9H-fluoren-9-ylmethyl-6-[5-cyano-1H-pyrrol-2-yl]-2,4,4-trimethyl-2H-3,1-benzoxazine-1(4H)-carboxylate was obtained as a white solid (0.4g, 97%).

A solution of 9H-fluoren-9-ylmethyl-6-[5-cyano-1H-pyrrol-2-yl]-2,4,4-trimethyl-2H-3,1-benzoxazine-1(4H)-carboxylate (0.1g, 0.2 mmol) in 20% piperidine in DMF (5 mL) was stirred at room temperature under nitrogen for 10 minutes. The mixture was poured into a saturated ammonium sulfate solution (30 mL) and extracted with diethyl ether (3x30 mL). The combined organic layers were dried (MgSO<sub>4</sub>) and evaporated. The residue was purified by a silica gel column chromatography(hexane:ethyl acetate/3:1) to afford the title compound as a white solid (0.03g, 56%): mp 201-202 °C; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 12.27 (s, 1H), 7.44 (d, 1H, *J* = 2.1 Hz), 7.32 (dd, 1H, *J* = 8.3, 1.1 Hz), 6.92 (dd, 1H, *J* = 4.1, 2.6 Hz), 6.57 (d, 1H, *J*

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= 8.3 Hz), 6.50 (dd, 1H,  $J = 4.2, 2.6$  Hz), 6.40 (s, 1H), 4.77-4.80 (m, 1H), 1.51 (s, 3H), 1.46 (s, 3H), 1.27 (d, 3H,  $J = 5.7$  Hz); MS (ESI)  $m/z$  266 [M-H]<sup>-</sup>.

#### EXAMPLE 21

5 9H-Fluoren-9-ylmethyl 6-(5-cyano-1-methyl-1H-pyrrol-2-yl)-2,4,4-trimethyl-2H-3,1-benzoxazine-1(4H)-carboxylate

A mixture of 9H-fluoren-9-ylmethyl-6-[5-cyano-1H-pyrrol-2-yl]-2,4,4-trimethyl-2H-3,1-benzoxazine-1(4H)-carboxylate (0.4g, 0.8 mmol) and potassium carbonate (1.5g) in anhydrous DMF was treated at ambient temperature under a  
10 blanket of nitrogen with iodomethane (1.5 mL, excess). The mixture was stirred for 30 minutes. A saturated ammonium sulfate solution (50 mL) and ethyl acetate (50 mL) was added. The organic layer was separated and aqueous layer was extracted with ethyl acetate (3x15 mL). The combined organic layers were dried (MgSO<sub>4</sub>),  
15 evaporated to yield the title compound as a white solid (0.35g, 87%): mp 63-64 °C; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 7.90 (m, 2H), 7.62 (d, 1H,  $J = 7.7$  Hz), 7.58 (d, 1H,  $J = 7.7$  Hz), 7.40 (m, 2H), 7.29-7.32 (m, 3H), 7.14 (dd, 1H,  $J = 8.1, 1.9$  Hz), 7.01-7.04 (m, 2H), 6.33 (d, 1H,  $J = 4.3$  Hz), 5.31 (q, 1H,  $J = 5.8$  Hz), 4.88 (dd, 1H,  $J = 10.8, 5.0$  Hz), 4.65 (dd, 1H,  $J = 10.8, 4.6$  Hz), 4.34 (t, 1H,  $J = 4.6$  Hz), 3.71 (s, 3H), 1.52 (s, 3H), 1.21 (s, 3H), 1.06 (d, 3H,  $J = 5.4$  Hz); MS (ESI)  $m/z$  504 [M+H]<sup>+</sup>.

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#### EXAMPLE 22

1-Methyl-5-(2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazin-6-yl)-1H-pyrrole-2-carbonitrile

Prepared according to the procedure for Example 20 from 9H-fluoren-9-ylmethyl 6-(5-cyano-1-methyl-1H-pyrrol-2-yl)-2,4,4-trimethyl-2H-3,1-benzoxazine-1(4H)-carboxylate (0.3g, 0.6 mmol) and 20% piperidine in DMF. A white solid  
25 (0.07g, 42%): mp 195-196 °C; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 7.16 (d, 1H,  $J = 2.8$  Hz), 7.08 (dd, 1H,  $J = 8.1, 1.9$  Hz), 6.98 (d, 1H,  $J = 4.1$  Hz), 6.63 (d, 1H,  $J = 8.5$  Hz), 6.61 (s,



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1H), 6.20 (d, 1H,  $J = 4.1$  Hz), 4.79-4.81 (m, 1H), 3.67 (s, 3H), 1.49 (s, 3H), 1.45 (s, 3H), 1.27 (d, 3H,  $J = 5.5$  Hz); MS (ESI)  $m/z$  282  $[M+H]^+$ .

### EXAMPLE 23

5    5-(2-Methylspiro[4H-3,1-benzoxazine-4,1'-cyclopentane]-6-yl)-4 methyl-2-thiophenecarbonitrile

2-Methyl-6-bromospiro[4H-3,1-benzoxazine-4,1'-cyclopentane] was prepared using the same procedure as for 6-bromo-2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazine in Example 13.

10        The title compound was prepared according to the procedure for Example 13 from 2-methyl-6-bromospiro[4H-3,1-benzoxazine-4,1'-cyclopentane] and 5-bromo-4-methyl-2-thiophenecarbonitrile. A yellowish solid: mp 58-60 °C;  $^1\text{H-NMR}$  (DMSO- $d_6$ )  $\delta$  7.79 (s, 1H), 7.16 (d, 1H,  $J = 1.9$  Hz), 7.12 (dd, 1H,  $J = 8.4, 2.2$  Hz), 6.66 (s, 1H), 6.63 (d, 1H,  $J = 8.4$  Hz), 4.75 (m, 1H), 2.26 (s, 3H), 2.14 (m, 1H), 1.87 (m, 1H),  
15    1.4-1.7 (m, 6H), 1.32 (d, 3H,  $J = 5.5$  Hz).

### EXAMPLE 24

4-(2-Methylspiro[2H-3,1-benzoxazine-4,1'-cyclopentane]-6-yl)-2-thiophenecarbonitrile

20        Prepared according to the procedure for Example 13 from 2-methyl-6-bromospiro[4H-3,1-benzoxazine-4,1'-cyclopentane] and 4-bromo-2-thiophenecarbonitrile. An off-white solid: mp 103-104 °C.

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## EXAMPLE 25

**5-(2-methylspiro[2H-3,1-benzoxazine-4,1'-cyclohexane]-6-yl)-4-methyl-2-thiophenecarbonitrile**

2-Methyl-6-bromospiro[2H-3,1-benzoxazine-4,1'-cyclohexane] was prepared  
5 using the same procedure as for 6-bromo-2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazine in Example 13.

The title compound was prepared according to the procedure for Example 13 from 2-methyl-6-bromospiro[4H-3,1-benzoxazine-4,1'-cyclohexane] and 5-bromo-4-methyl-2-thiophenecarbonitrile. A brown solid: <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 7.78 (s, 1H),  
10 7.17 (d, 1H, *J* = 1.8 Hz), 7.14 (dd, 1H, *J* = 8.4, 2.2 Hz), 6.64 (s, 1H), 6.63 (d, 1H, *J* = 8.2 Hz), 4.74 (m, 1H), 2.26 (s, 3H), 2.14 (m, 1H), 1.87 (m, 1H), 1.4-1.7 (m, 8H), 1.31 (d, 3H, *J* = 5.3 Hz); MS (ESI) *m/z* 337 [M-H]<sup>-</sup>.

## EXAMPLE 26

15 **4-(2-Methylspiro[2H-3,1-benzoxazine-4,1'-cyclohexane]-6-yl)2-thiophenecarbonitrile**

Prepared according to the procedure for Example 13 from 2-methyl-6-bromospiro[4H-3,1-benzoxazine-4,1'-cyclohexane] and 4-bromo-2-thiophenecarbonitrile. A brown solid: mp 111-112 °C; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 8.42  
20 (s, 1H), 8.14 (s, 1H), 7.47 (s, 1H), 7.35 (dd, 1H, *J* = 8.3, 1.1 Hz), 6.58 (d, 1H, *J* = 8.4 Hz), 6.38 (s, 1H), 4.72 (m, 1H), 1.92-2.16 (m, 2H), 1.35-1.75 (m, 8H), 1.31 (d, 3H, *J* = 5.3 Hz); MS (ESI) *m/z* 325 [M+H]<sup>+</sup>.

## EXAMPLE 27

25 **6-(3-Fluorophenyl)-2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazine**

Prepared according to the coupling procedure for Example 17 from 3-fluorophenyl boronic acid and 6-bromo-2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazine. A yellow solid: mp 139-140°C; <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ 7.40-7.19 (m, 6H),

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7.01-6.94 (m, 1H), 6.72 (d, 1H,  $J = 8.24$  Hz), 4.90 (q, 1H,  $J = 5.48$  Hz), 1.62 (s, 3H), 1.59 (s, 3H), 1.46 (d, 3H,  $J = 5.5$  Hz); MS (ES)  $m/z$  272 ( $[M+H]^+$ ).

**EXAMPLE 28****5    6-(3-Chlorophenyl)-2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazine**

Prepared according to the coupling procedure for Example 17 from 3-chlorophenyl boronic acid and 6-bromo-2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazine. A orange solid: mp 144-146 °C;  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$  7.50 (t, 1H,  $J = 1.78$  Hz), 7.40 (dt, 1H,  $J = 7.61, 1.45$  Hz), 7.33 (t, 1H,  $J = 7.76$  Hz), 7.29-7.22 (m, 4H) 6.72 (d, 1H,  $J = 8.24$  Hz), 4.90 (q, 1H,  $J = 5.45$  Hz), 1.62 (s, 3H), 1.59 (s, 3H), 1.46 (d, 3H,  $J = 5.5$  Hz); MS (ES)  $m/z$  288/290 ( $[M + H]^+$ ).

**EXAMPLE 29****6-(3-Chloro-4-fluorophenyl)-2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazine**

15        A mixture of 6-bromo-2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazine (3.0 g, 11.7 mmol), 3-chloro-4-fluorobenzeneboronic acid (3.1 g, 17.6 mmol), tetrakis(triphenylphosphine)palladium (0) (0.67 g, 0.59 mmol), and sodium carbonate (3.72 g, 35.1 mmol) in DME (80 mL) and water (40 mL) was subject to a blanket of nitrogen flow for 15 minutes at 50°C and then was heated at 85°C under nitrogen for 1  
20    hour. The reaction was cooled to room temperature and ethyl acetate (200 mL) was added. The organic layer was washed twice with aqueous ammonium chloride (50 mL) and once with brine (50 mL), dried over sodium sulfate and concentrated to a yellow solid. The solid was triturated with ether (25 mL) and hexane (25 mL), collected on a filter, and dried to give 6-(3-chloro-4-fluorophenyl)-2,4,4-trimethyl-1,4-  
25    dihydro-2H-3,1-benzoxazine (1.87 g, 35%) as a yellow solid: mp 173-175°C;  $^1\text{H-NMR}$  ( $\text{DMSO-}d_6$ )  $\delta$  7.79 (dd, 1H,  $J = 6.84, 2.14$  Hz), 7.60-7.57 (m, 1H), 7.43-7.39 (m, 2H), 7.29 (dd, 1H,  $J = 8.12, 2.14$  Hz), 6.62 (d, 1H,  $J = 8.54$  Hz), 6.40 (s, 1H),

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4.79 (q, 1H,  $J = 5.55$  Hz), 1.53 (s, 3H), 1.48 (s, 3H), 1.28 (d, 3H,  $J = 5.55$  Hz); MS (ES)  $m/z$  306/308( $[M + H]^+$ ).

### EXAMPLE 30

5    **2-Fluoro-5-(2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazin-6-yl)benzonitrile**

Prepared according to the coupling procedure for Example 13 from 6-bromo-2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazine and 5-bromo-2-fluorobenzonitrile. A white solid: mp 184-186°C;  $^1\text{H-NMR}$  (DMSO- $d_6$ )  $\delta$  8.17 (dd, 1H,  $J = 5.86, 3.42$  Hz), 8.00-7.98 (m, 1H), 7.52 (t, 1H,  $J = 9.28$  Hz), 7.45 (d, 1H,  $J = 1.95$  Hz), 7.34 (dd, 1H,  $J = 8.30, 1.95$  Hz), 6.63 (d, 1H,  $J = 8.3$  Hz), 6.45 (d, 1H), 4.8 (q, 1H,  $J = 5.37$  Hz), 1.53 (s, 3H), 1.49 (s, 3H), 1.29 (d, 3H,  $J = 5.37$  Hz); MS (ES)  $m/z$  297 ( $[M + H]^+$ ).

### EXAMPLE 31

15    **4-(2,4,4-Trimethyl-1,4-dihydro-2H-3,1-benzoxazin-6-yl)-2-furonitrile**

Prepared using the coupling procedure for Example 13 from 6-bromo-2,4,4-trimethyl-1,4-dihydro-2H-3,1-benzoxazine and 4-bromo-2-cyanofuran. A light brown solid: mp 116-118 °C;  $^1\text{H-NMR}$  (DMSO- $d_6$ )  $\delta$  8.43 (s, 1H), 8.06 (s, 1H), 7.38 (d, 1H,  $J = 1.98$  Hz), 7.25 (dd, 1H,  $J = 7.93, 1.98$  Hz) 6.58 (d, 1H,  $J = 7.93$  Hz), 6.37 (s, 1H), 4.77 (q, 1H,  $J = 5.55$  Hz), 1.50 (s, 3H), 1.46 (s, 3H), 1.27 (d, 3H,  $J = 5.55$  Hz); MS (ES)  $m/z$  269 ( $[M+H]^+$ ); Anal. Calc. For  $\text{C}_{16}\text{H}_{16}\text{N}_2\text{O}_2$ : C, 71.62; H, 6.01, N, 10.44. Found: C, 71.55; H, 6.26, N, 10.17.

### EXAMPLE 32

25    **3-[4,4-Dimethyl-2-(trifluoromethyl)-1,4-dihydro-2H-3,1-benzoxazin-6-yl]-5-fluorobenzonitrile**

A mixture of 2-(2-amino-5-bromo-phenyl)-propan-2-ol (5 g, 21.7 mmol), trifluoroacetaldehyde methyl hemiacetal (5 mL), and magnesium sulfate (10 g) in toluene (75 mL) was heated at 80°C under nitrogen. After disappearance of the

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starting material, the reaction was filtered through a pad of silica gel and the filtrate dried over sodium sulfate and concentrated to give 6-bromo-4,4-dimethyl-2-(trifluoromethyl)-1,4-dihydro-2H-3,1-benzoxazine as an amorphous solid: <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 7.31 (d, 1H, *J* = 1.99 Hz), 7.19 (dd, 1H, *J* = 8.73, 2.38 Hz), 6.88 (s, 1H), 6.74 (d, 1H, *J* = 8.33 Hz), 5.32 (m, 1H), 1.52 (s, 6H).

A mixture of 6-bromo-4,4-dimethyl-2-(trifluoromethyl)-1,4-dihydro-2H-3,1-benzoxazine (0.5 g, 1.61 mmol), 3-cyano-5-fluorobenzeneboronic acid (0.39 g, 1.8 mmol), tetrakis(triphenylphosphine)-palladium (0) (0.2 g, 0.161 mmol), and sodium carbonate (0.5 g, 4.83 mmol) in DME (50 mL) and water (25 mL) was subject to a blanket of nitrogen flow for 15 minutes at 50°C and then was heated at 85°C under nitrogen for 1 hour. The reaction mixture was cooled to room temperature and ethyl acetate (200 mL) was added. The organic layer was washed twice with aqueous ammonium chloride (20 mL) and once with brine (20 mL), dried over sodium sulfate and concentrated to a yellow solid. Purification via flash column chromatography (silica gel, 5% ethyl acetate/hexane) gave 3-[4,4-dimethyl-2-(trifluoromethyl)-1,4-dihydro-2H-3,1-benzoxazin-6-yl]-5-fluorobenzonitrile (0.396 g, 73%) as a white solid: mp 102-103°C; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 8.07 (s, 1H), 7.91 (d, 1H, *J* = 10.7 Hz), 7.7 (d, 1H, *J* = 10.7 Hz), 7.62 (d, 1H, *J* = 1.98 Hz), 7.53 (dd, 1H, *J* = 8.33, 1.98 Hz), 7.05 (s, 1H), 6.87 (d, 1H, *J* = 8.33 Hz), 5.39-5.37 (m, 1H), 1.61 (s, 3H), 1.59 (s, 3H); MS (ES) *m/z* 349 ([M-H]<sup>+</sup>).

### EXAMPLE 33

#### 4-[4,4-Dimethyl-2-(trifluoromethyl)-1,4-dihydro-2H-3,1-benzoxazin-6-yl]thiophene-2-carbonitrile

Prepared using the coupling procedure for Example 13 starting with 6-bromo-4,4-dimethyl-2-(trifluoromethyl)-1,4-dihydro-2H-3,1-benzoxazine and 4-bromo-2-thiophenecarbonitrile. A yellow solid: <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 8.43 (s, 1H), 8.19 (s, 1H), 7.54 (d, 1H, *J* = 1.98 Hz), 7.46 (dd, 1H, *J* = 8.33, 1.98 Hz), 6.90 (s, 1H), 6.83 (d,

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1H,  $J = 8.33$  Hz), 5.35 (d, 1H,  $J = 3.57$  Hz), 1.59 (s, 3H), 1.57 (s, 3H); MS (ES)  $m/z$  337 ([M-H]).

#### EXAMPLE 34

5 4-[1-Acetyl-4,4-dimethyl-2-(trifluoromethyl)-1,4-dihydro-2H-3,1-benzoxazin-6-yl]thiophene-2-carbonitrile

4-[4,4-dimethyl-2-(trifluoromethyl)-1,4-dihydro-2H-3,1-benzoxazin-6-yl]thiophene-2-carbonitrile (0.25 g, 0.74 mmol) was dissolved in DMF (10 mL), NaH (0.09 g, 2.22 mmol) was added and the mixture was stirred for 30 minutes prior to the  
10 addition of acetyl chloride (0.079 mL, 1.11 mmol). Upon disappearance of the starting material the reaction mixture was poured into brine (100 mL) and the product extracted with ether (150 mL), dried over sodium sulfate and concentrated. The residue was purified by a flash column chromatography (silica gel, 25% ethyl acetate/hexane) to yield 4-[1-acetyl-4,4-dimethyl-2-(trifluoromethyl)-1,4-dihydro-2H-  
15 3,1-benzoxazin-6-yl]thiophene-2-carbonitrile (0.1 g, 36%) as an amorphous solid:  $^1\text{H}$ -NMR (DMSO- $d_6$ )  $\delta$  8.58 (s, 1H), 8.5 (s, 1H), 7.83-7.80 (m, 2H), 7.64 (d, 1H,  $J = 8.74$  Hz), 6.42 (q, 1H,  $J = 10$  Hz), 2.24 (s, 3H), 1.73 (s, 3H), 1.42 (s, 3H).

#### EXAMPLE 35

20 (6(5-Cyanothien-3-yl)-4,4-dimethyl-2-(trifluoromethyl)-2H-3,1-benzoxazin-1(4H)-yl)methyl pivalate

4-[4,4-dimethyl-2-(trifluoromethyl)-1,4-dihydro-2H-3,1-benzoxazin-6-yl]thiophene-2-carbonitrile (0.30 g, 0.89 mmol) was dissolved in DMF (10 mL), NaH (0.065 g, 2.7 mmol) was added and the mixture stirred for 30 minutes prior to the  
25 addition of *tert*-butyl chloroacetate (0.19 mL, 1.3 mmol). Upon disappearance of the starting material the reaction mixture was poured into brine (100 mL) and the product extracted with ether (150 mL), dried over sodium sulfate and concentrated. Flash column chromatography (silica gel, 8% ethyl acetate/hexane) gave (6(5-cyanothien-3-

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yl)-4,4-dimethyl-2-(trifluoromethyl)-2H-3,1-benzoxazin-1(4H)-yl)methyl pivalate  
(0.127 g, 32%) as an amorphous solid:  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$  7.82 (s, 1H), 7.58 (s, 1H),  
7.41 (dd, 1H,  $J = 8.33, 1.98$  Hz), 7.25 (d, 1H,  $J = 1.98$  Hz) 7.10 (d, 1H,  $J = 8.33$  Hz),  
6.79 (d, 1H,  $J = 8.33$  Hz), 5.66 (d, 1H,  $J = 11.9$  Hz), 5.38 (d, 1H,  $J = 11.9$  Hz), 1.72  
5 (s, 3H), 1.55 (s, 3H), 1.18 (s, 9H).

**EXAMPLE 36****3-Fluoro-5-(2,2,4,4-tetramethyl-1,4-dihydro-2H-3,1-benzoxazin-6-yl)benzonitrile**

Prepared using the coupling procedure for Example 13 starting with 6-bromo-  
10 2,2,4,4-tetramethyl-1,4-dihydro-2H-3,1-benzoxazine and 3-bromo-5-  
fluorobenzonitrile. A white solid:  $^1\text{H-NMR}$  ( $\text{DMSO}-d_6$ )  $\delta$  8.03 (s, 1H), 7.87 (d, 1H,  $J$   
= 10.7 Hz), 7.65 (d, 1H,  $J = 10.7$  Hz), 7.57 (d, 1H,  $J = 1.98$  Hz) 7.45 (dd, 1H,  $J$  =  
8.33, 1.98 Hz), 6.65 (d, 1H,  $J = 8.33$  Hz), 6.42 (s, 1H), 1.52 (s, 6H), 1.34 (s, 6H);  
MS (ES)  $m/z$  311 ( $[\text{M}+\text{H}]^+$ ).

15

**EXAMPLE 37****4-(2,2,4,4-Tetramethyl-1,4-dihydro-2H-3,1-benzoxazin-6-yl)thiophene-2-carbonitrile**

Prepared using the coupling procedure for Example 13 starting with 6-bromo-  
20 2,2,4,4-tetramethyl-1,4-dihydro-2H-3,1-benzoxazine and 4-bromo-2-cyanothiophene.  
An amorphous orange solid:  $^1\text{H-NMR}$  ( $\text{DMSO}-d_6$ )  $\delta$  8.4 (d, 1H,  $J = 1.69$  Hz), 8.14 (d,  
1H,  $J = 1.69$  Hz), 7.49 (d, 1H,  $J = 2.20$  Hz), 7.38 (dd, 1H,  $J = 8.43, 2.02$  Hz), 6.61  
(d, 1H,  $J = 8.43$  Hz), 6.26 (s, 1H), 1.49 (s, 6H), 1.32 (s, 6H); MS (ES)  $m/z$  299  
( $[\text{M}+\text{H}]^+$ ).

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**EXAMPLE 38****5-[4,4-Dimethyl-2-(trifluoromethyl)-1,4-dihydro-2H-3,1-benzoxazin-6-yl]-4-methylthiophene-2-carbonitrile**

Prepared according to the coupling procedure for Example 13 from 6-bromo-  
5 4,4-dimethyl-2-(trifluoromethyl)-1,4-dihydro-2H-3,1-benzoxazine and 5-bromo-4-methyl-2-thiophene-carbonitrile. A yellowish solid: mp 97-98 °C; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 7.82 (s, 1H), 7.26 (d, 1H, *J* = 1.7 Hz), 7.21 (dd, 1H, *J* = 8.1, 2.1 Hz), 7.16 (s, 1H), 6.88 (d, 1H, *J* = 8.1 Hz), 5.40 (m, 1H), 2.27 (s, 3H), 1.56 (s, 3H), 1.55 (s, 3H); MS (ESI) *m/z* 351 [M-H]<sup>+</sup>.

10

**EXAMPLE 39****6-(3-Chlorophenyl)-4,4-dimethyl-2-(trifluoromethyl)-1,4-dihydro-2H-3,1-benzoxazine**

Prepared according to the coupling procedure for Example 17 from 6-bromo-  
15 4,4-dimethyl-2-(trifluoromethyl)-1,4-dihydro-2H-3,1-benzoxazine and 3-chlorophenyl boronic acid. A yellowish solid: mp 108-109 °C; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 7.69 (t, 1H, *J* = 1.7 Hz), 7.59 (d, 1H, *J* = 7.8 Hz), 7.35-7.50 (m, 3H), 7.32 (dt, 1H, *J* = 8.1, 1.1 Hz), 6.91 (s, 1H), 6.87 (d, 1H, *J* = 8.4 Hz), 5.35 (m, 1H), 1.60 (s, 3H), 1.59 (s, 3H); MS (ESI) *m/z* 340 [M-H]<sup>+</sup>.

20

**EXAMPLE 40****6-(3-Fluorophenyl)-4,4-dimethyl-3,4-dihydroquinolin-2(1H)-one**

Prepared by coupling 3-fluorophenylboronic acid with an equivalent amount of  
6-bromo-4,4-dimethyl-3,4-dihydro-1H-quinolin-2-one using a catalytic amount of  
25 tetrakis(triphenylphosphine)palladium(0) with overnight refluxing in toluene containing an equivalent amount of potassium carbonate dissolved in water. Work up in the usual manner followed by recrystallization from ethanol gave a gray solid: mp 190-192° C.



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<sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 10.27 (s, 1H), 7.57 (s, 1H), 7.50 (m, 4H), 7.15 (m, 1H), 6.95 (d, 1H *J* = 8.2Hz), 2.39 (s, 2H), 1.29 (s, 6H); MS (APCI (-)) *m/z* 268 [M-H]<sup>-</sup> Anal. Calc. For C<sub>17</sub>H<sub>16</sub>FNO 0.25 H<sub>2</sub>O: C, 74.57, H, 6.07, N, 5.12. Found: C, 74.86, H, 5.97, N 5.06.

5

**EXAMPLE 41****3-(4,4-Dimethyl-2-oxo-1,2,3,4-tetrahydroquinolin-6-yl)-5-fluorobenzonitrile**

6-bromo-4,4-quinolin-2-one was allowed to couple with an equivalent of bis(pinacolato)diboron in refluxing DMF containing an equivalent of sodium carbonate dissolved in a minimal amount of water and a catalytic quantity of  
10 tetrakis(triphenylphosphine)palladium(0). After refluxing overnight an equivalent of 3-bromo-5-fluorobenzonitrile was added. Another equivalent of sodium carbonate was then added, followed by an additional amount of the same catalyst. After several hours of reflux, the reaction mixture was filtered and taken to dryness *in vacuo*. The  
15 residue was extracted into ethyl acetate and the solution was dried over magnesium sulfate, filtered and the filtrate was again roto-evaporated to give a solid residue. Recrystallization from ethanol afforded the title compound as a gray solid: mp 249-250° C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 10.32 (s, 1H), 8.09 (t, 1H, *J* = 1.6Hz), 7.93 (d, 1H, *J* = 12.7 Hz) 7.77, (d, 1H, *J* = 8.1Hz), 7.70 (s, 1H), 7.61(d, 1H *J* = 8.9Hz), 6.95 (d,  
20 1H, *J* = 8.3Hz), 2.40 (s, 2H), 1.30 (s, 6H); MS (APCI (-)) *m/z* 293 [M-H]<sup>-</sup> Anal. Calc. For C<sub>18</sub>H<sub>15</sub>FN<sub>2</sub>O 1.5H<sub>2</sub>O : C, 67.28, H, 5.65, N, 8.72. Found: C, 67.36, H, 4.90, N 8.44..

**EXAMPLE 42****3-(4,4-Dimethyl-2-oxo-1,2,3,4-tetrahydroquinolin-6-yl)benzonitrile**

Prepared by coupling 3-cyanophenylboronic acid with an equivalent amount of 6-bromo-4,4-dimethyl-3,4-dihydro-1H-quinolin-2-one using a catalytic amount of tetrakis(triphenylphosphine)palladium(0) as a catalyst with overnight refluxing in toluene containing an equivalent amount of potassium carbonate dissolved in water in

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the usual manner followed by recrystallization from ethanol gave a gray solid: mp. 190 – 192 °C; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 10.29 (s, 1H), 8.17 (s, 1H), 8.00 (d, 1H, J = 7.9 Hz), 7.77 (d, 1H, J = 6.4 Hz), 7.65 (m, 2H), 7.55 (d, 1H, J = 8.2 Hz), 6.97 (d, 1H, J = 8.3 Hz), 2.40 (s, 2H), 1.30 (s, 6H); MS (APCI (-)) *m/z* 275 [M-H]<sup>-</sup> Anal. Calc. For C<sub>18</sub>H<sub>16</sub>N<sub>2</sub>O 0.25H<sub>2</sub>O: C, 75.77, H, 6.01, N, 9.82. Found: C, 75.45, H, 5.65, N 9.20.

### EXAMPLE 43

#### 6-(3-Chlorophenyl)-4,4-dimethyl-3,4-dihydroquinoline-2(1H)-thione

Prepared by heating under reflux overnight a mixture of 6-(3-chlorophenyl)-  
4,4-dimethyl-3,4-dihydroquinoline-2(1H)-one and an equal weight of phosphorus  
pentasulfide in pyridine was stirred. Removal of the pyridine *in vacuo* followed by  
treating the residue with 6N hydrochloric acid and recrystallization of the residue in  
ethanol gave the product as a yellow solid: mp. 197 - 198 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ  
12.34 (s, 1H), 7.75 (m, 1H), 7.64 (m, 2H), 7.57 (dd, 1H, J = 9.3 and 2.1 Hz), 7.48 (t,  
1H, J = 7.7 Hz), 7.41 (m, 1H), 7.20 (d, 1H, J = 8.3 Hz), 3.34 (s, 2H), 1.26 (s, 6H); MS  
(APCI (-)) *m/z* 300 [M-H]<sup>-</sup> Anal. Calc. For C<sub>17</sub>H<sub>16</sub>ClNS: C, 67.65, H, 5.34, N, 4.64.  
Found: C, 67.77, H, 5.57, N 4.54.

### EXAMPLE 44

#### 6-(3-Fluorophenyl)-4,4-dimethyl-3,4-dihydroquinoline-2(1H)-thione

Prepared by heating under reflux overnight a mixture of 6-(3-fluorophenyl)-  
4,4-dimethyl-3,4-dihydroquinoline-2(1H)-one and an equal weight of phosphorus  
pentasulfide was stirred in pyridine. Workup as in the previous example gave a yellow  
solid, mp 209-211 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 12.34 (s, 1H), 7.65 (d, 1H, J = 2.2 Hz )  
7.57 (dd, 1H, J<sub>1</sub> = 8.24 J<sub>2</sub> = 2.2 Hz), 7.51 (m, 3H), 7.18 (m, 2H), 2.84 (s, 2H), 1.26 (s, 6H).  
MS *m/z* 284 [M-H]<sup>-</sup> Anal. Calc. For C<sub>17</sub>H<sub>16</sub>FINs: C, 71.55, H, 5.65, N, 4.91. Found:  
C, 71.18, H, 5.59, N 4.82.

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**EXAMPLE 45****3-(4,4-Dimethyl-2-thioxo-1,2,3,4-tetrahydroquinolin-6-yl)benzonitrile**

Prepared by heating under reflux overnight a stirred mixture of 3-(4,4-dimethyl-2-oxo-1,2,3,4-tetrahydroquinolin-6-yl)benzonitrile and an equal weight of phosphorus pentasulfide in pyridine and workup as in the previous example gave a yellow solid: mp 220-223 °C dec. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 12.35 (s, 1H), 8.21 (s, 1H), 8.10 (d, 1H, *J* = 6.0 Hz), 7.80 (d, 1H, *J* = 7.9 Hz), 7.72 (s, 1H), 7.65 (m, 2H), 7.21 (d, 1H, *J* = 8.4 Hz), 2.85 (s, 2H), 1.27 (s, 6H); MS (APCI (-)) *m/z* 291 [M-H]<sup>-</sup> Anal. Calc. For C<sub>18</sub>H<sub>16</sub>N<sub>2</sub>S · 3H<sub>2</sub>O: C, 62.40, H, 6.40, N, 8.09. Found: C, 62.12, H, 4.88, N 7.77.

**EXAMPLE 46****3-(4,4-Dimethyl-2-thioxo-1,2,3,4-tetrahydroquinolin-6-yl)-5-fluorobenzonitrile**

Prepared by heating under reflux overnight a stirred mixture of 3-(4,4-dimethyl-2-oxo-1,2,3,4-tetrahydroquinolin-6-yl)-5-fluorobenzonitrile and an equal weight of phosphorus pentasulfide in pyridine and workup as in the previous example gave a yellow solid: mp. 240-242 °C dec.; <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ 12.37 (s, 1H), 8.13 (s, 1H), 7.98 (dt, 1H, *J* = 10.4 and 5.4 Hz), 7.81 (d, 1H, *J* = 8.4 Hz), 7.7 (d, 1H, *J* = 1.8 Hz), 7.68 (dd, 1H, *J* = 8.3 and 1.9 Hz), 7.22 (d, 1H, *J* = 8.3 Hz), 2.85 (s, 2H), 1.27 (s, 6H); MS *m/z* 309 [M-H]<sup>-</sup> Anal. Calc. For C<sub>18</sub>H<sub>15</sub>FN<sub>2</sub>S · 0.10 H<sub>2</sub>O: C, 69.25, H, 4.91, N, 8.97. Found: C, 69.15, H, 4.74, N 8.75.

**EXAMPLE 47****6-(3-Fluoro-phenyl)-2,4,4-trimethyl-1,2,3,4-tetrahydro-quinoline****6-Bromo-1-(4-methoxy-benzyl)-4,4-dimethyl-3,4-dihydro-1H-quinolin-2-one.**

To a solution of 6-bromo-4,4-dimethyl-3,4-dihydro-1H-quinolin-2-one

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(0.5g, 1.97mmol) in THF (25mL) was added 60% NaH (0.12g, 2.95mmol) suspended in mineral oil. The resulting reaction mixture was stirred at room temperature for 30 min., 4-methoxybenzyl chloride (0.34g, 2.17) added, and heated under reflux for 20 h. The reaction was cooled to room temperature then quenched slowly with water. After  
5 extraction with ethyl acetate, the organic layer was dried (MgSO<sub>4</sub>), evaporated and the residue purified by chromatography (SiO<sub>2</sub> 3:7 ethyl acetate/hexane). The white crystalline product was obtained (0.35g, 48%); mp 118-119 °C, <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>) δ 1.23 (s, 6H), 2.59 (s, 2H), 3.70 (s, 3H), 3.72 (s, 1H), 4.41 (d, 1H, *J* = 5.86Hz), 5.09 (s, 1H), 6.87 (m, 2H), 7.01 (d, 1H, *J* = 8.78Hz), 7.17 (d, 1H, *J* = 8.98Hz), 7.23 (d,  
10 1H, *J* = 8.79), 7.34 (dd, 1H, *J* = 6.59 and 2.2 Hz), 7.43 (d, 1H, *J* = 2.2Hz); MS (APCI (+)) [M+H]<sup>+</sup> = 374/376.

6-(3-Fluoro-phenyl)-1-(4-methoxy-benzyl)-4,4-dimethyl-3,4-dihydro-1H-quinolin-2-one.

A mixture of 6-Bromo-1-(4-methoxy-benzyl)-4,4-dimethyl-3,4-dihydro-1H-  
15 quinolin-2-one (2.9g, 7.75mmol) in ethylene glycol dimethyl ether (50mL), K<sub>2</sub>CO<sub>3</sub> (1.18 g, 8.53mmol) in H<sub>2</sub>O (5.0mL), and a catalytic amount of tetrakis(triphenylphosphine) palladium was heated under reflux overnight. After cooling to room temperature, the mixture was extracted with ethyl acetate and the organic phase was washed with NaHCO<sub>3</sub> solution, brine, dried over MgSO<sub>4</sub>, concentrated, and  
20 crystallized from ethanol to obtain the product as a white crystalline material (1.8 g, 60%); mp 159 – 162 °C, <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>) δ 1.32 (s, 6H), 2.63 (s, 2H), 3.70 (s, 3H), 5.15 (s, 2H), 6.89 (d, 2H, *J* = 11.72Hz), 7.14 (m, 2H), 7.22 (d, 2H, *J* = 8.8Hz), 7.53 (m, 4H), 7.60 (d, 1H, *J* = 2.2Hz); MS (APCI (+)) [M+H]<sup>+</sup> = 390.

6-Bromo-1-(4-methoxy-benzyl)-2,4,4-trimethyl-1,4-dihydro-quinoline.

25 To a solution of 6-(3-Fluoro-phenyl)-1-(4-methoxy-benzyl)-4,4-dimethyl-3,4-dihydro-1H-quinolin-2-one (1.16g, 2.99mmol) in THF (15mL) at room temperature was added a solution of 1.4 M MeMgBr (2.6mL, 12.56mmol) and the resulting reaction mixture was stirred for 6h. The solution was quenched with H<sub>2</sub>O, extracted

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with EtOAc, treated with ammonium chloride solution, washed with brine, dried over MgSO<sub>4</sub>, and concentrated. The product was purified by column chromatography using a 2:8 hexane/ethyl acetate mixture and used below.

To a solution of 6-(3-Fluoro-phenyl)-1-(4-methoxy-benzyl)-2,4,4-trimethyl-  
5 1,2,3,4-tetrahydroquinoline (0.15g, 0.39mmol) in EtOH (35mL) was added 10%Pd/C and hydrogenated for 10 h at 40 psi. The catalyst was removed by filtration through celite and the product purified by chromatography. A reddish liquid was obtained. <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>) δ 1.15 (d, 3H, *J* = 6.73Hz), 1.21 (s, 3H), 1.34(m, 4H), 1.60( dd, 1H, *J*<sub>1</sub> = 10.1, *J*<sub>2</sub> = 2.69, Hz), 3.40( m, 1H), 5.95( s, 1H), 6.55( d, 1H, *J* = 8.08Hz),  
10 7.01( m, 1H), 7.21(dd, 1H, *J*<sub>1</sub> = 5.78 *J*<sub>2</sub> = 2.02 Hz), 7.35(m, 4H); MS( FI Pos) [M+H]<sup>+</sup> = 270.

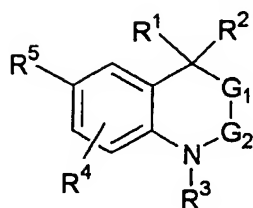
Still other compounds, including 5-(2,4,4-Trimethyl-1,2,3,4-tetrahydro-  
quinolin-6-yl)-1H-pyrrole-2-carbonitrile, 1-Methyl-5-(2,4,4-trimethyl-1,2,3,4-  
tetrahydro-quinolin-6-yl)-1H-pyrrole-2- carbonitrile, and 3-Fluoro-5-(2,4,4-trimethyl-  
15 1,2,3,4-tetrahydro-quinolin-6-yl)-benzonitrile may be prepared using the methods described above.

All publications cited in this specification are incorporated herein by reference herein. While the invention has been described with reference to a particularly  
20 preferred embodiment, it will be appreciated that modifications can be made without departing from the spirit of the invention. Such modifications are intended to fall within the scope of the appended claims.

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**What is Claimed:**

1. A compound of the formula:



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wherein:

$R^1$ ,  $R^2$  are independent substituents selected from the group which includes H,  $C_1$  to  $C_6$  alkyl, substituted  $C_1$  to  $C_6$  alkyl,  $C_2$  to  $C_6$  alkenyl, substituted  $C_2$  to  $C_6$  alkenyl,  $C_2$  to  $C_6$  alkynyl, substituted  $C_2$  to  $C_6$  alkynyl,  $C_3$  to  $C_8$  cycloalkyl, substituted  $C_3$  to  $C_8$  cycloalkyl, aryl, substituted aryl, heterocyclic, substituted heterocyclic,  $COR^A$ , or  $NR^B COR^A$ ;

or  $R^1$  and  $R^2$  are fused to form:

- an optionally substituted 3 to 8 membered spirocyclic alkyl ring;
- an optionally substituted 3 to 8 membered spirocyclic alkenyl; or
- an optionally substituted 3 to 8 membered heterocyclic ring containing one to three heteroatoms from the group including O, S and N; the spirocyclic rings of a), b) and c) being optionally substituted by from 1 to 4 groups selected from fluorine,  $C_1$  to  $C_6$  alkyl,  $C_1$  to  $C_6$  alkoxy,  $C_1$  to  $C_6$  thioalkyl,  $-CF_3$ ,  $-OH$ ,  $-CN$ ,  $NH_2$ ,  $-NH(C_1$  to  $C_6$  alkyl), or  $-N(C_1$  to  $C_6$  alkyl) $_2$ ;

$R^A$  is H,  $C_1$  to  $C_3$  alkyl, substituted  $C_1$  to  $C_3$  alkyl, aryl, substituted aryl,  $C_1$  to  $C_3$  alkoxy, substituted  $C_1$  to  $C_3$  alkoxy,  $C_1$  to  $C_3$  aminoalkyl, substituted  $C_1$  to  $C_3$  aminoalkyl,

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$R^B$  is H,  $C_1$  to  $C_3$  alkyl, substituted  $C_1$  to  $C_3$  alkyl,

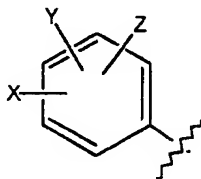
$R^3$  is H, OH,  $NH_2$ ,  $C_1$  to  $C_6$  alkyl, substituted  $C_1$  to  $C_6$  alkyl,  $C_3$  to  $C_6$  alkenyl, substituted  $C_1$  to  $C_6$  alkenyl, alkynyl, or substituted alkynyl,  $COR^C$ ,

$R^C$  is H,  $C_1$  to  $C_3$  alkyl, substituted  $C_1$  to  $C_3$  alkyl, aryl, substituted aryl,  $C_1$  to  $C_3$  alkoxy, substituted  $C_1$  to  $C_3$  alkoxy,  $C_1$  to  $C_3$  aminoalkyl, substituted  $C_1$  to  $C_3$  aminoalkyl,

$R^4$  is H, halogen, CN,  $NO_2$ ,  $C_1$  to  $C_6$  alkyl, substituted  $C_1$  to  $C_6$  alkyl, alkynyl, or substituted alkynyl,  $C_1$  to  $C_6$  alkoxy, substituted  $C_1$  to  $C_6$  alkoxy, amino,  $C_1$  to  $C_6$  aminoalkyl, substituted  $C_1$  to  $C_6$  aminoalkyl,

$R^5$  is selected from a) or b):

a)  $R^5$  is a trisubstituted benzene of the structure:



X is taken from the group including halogen, CN,  $C_1$  to  $C_3$  alkyl, substituted  $C_1$  to  $C_3$  alkyl, alkynyl, or substituted alkynyl,  $C_1$  to  $C_3$  alkoxy, substituted  $C_1$  to  $C_3$  alkoxy,  $C_1$  to  $C_3$  thioalkoxy, substituted  $C_1$  to  $C_3$  thioalkoxy, amino,  $C_1$  to  $C_3$  aminoalkyl, substituted  $C_1$  to  $C_3$  aminoalkyl,  $NO_2$ ,  $C_1$  to  $C_3$  perfluoroalkyl, 5 or 6 membered heterocyclic ring containing 1 to 3 heteroatoms,  $COR^D$ ,  $OCOR^D$ , or  $NR^E COR^D$ ;

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$R^D$  is H,  $C_1$  to  $C_3$  alkyl, substituted  $C_1$  to  $C_3$  alkyl, aryl, substituted aryl,  $C_1$  to  $C_3$  alkoxy, substituted  $C_1$  to  $C_3$  alkoxy,  $C_1$  to  $C_3$  aminoalkyl, or substituted  $C_1$  to  $C_3$  aminoalkyl;

$R^E$  is H,  $C_1$  to  $C_3$  alkyl, substituted  $C_1$  to  $C_3$  alkyl;

Y and Z are independent substituents taken from the group including H, halogen, CN,  $NO_2$ , amino, aminoalkyl,  $C_1$  to  $C_3$  alkoxy,  $C_1$  to  $C_3$  alkyl, or  $C_1$  to  $C_3$  thioalkoxy; or

b)  $R^5$  is a five or six membered ring with 1, 2, or 3 heteroatoms selected from O, S, SO,  $SO_2$  or  $NR^6$  and containing one or two independent substituents selected from H, halogen, CN,  $NO_2$ , amino, and  $C_1$  to  $C_3$  alkyl,  $C_1$  to  $C_3$  alkoxy,  $C_1$  to  $C_3$  aminoalkyl,  $COR^F$ , or  $NR^G COR^F$ ;

$R^F$  is H,  $C_1$  to  $C_3$  alkyl, substituted  $C_1$  to  $C_3$  alkyl, aryl, substituted aryl,  $C_1$  to  $C_3$  alkoxy, substituted  $C_1$  to  $C_3$  alkoxy,  $C_1$  to  $C_3$  aminoalkyl, or substituted  $C_1$  to  $C_3$  aminoalkyl;

$R^G$  is H,  $C_1$  to  $C_3$  alkyl, or substituted  $C_1$  to  $C_3$  alkyl;

$R^6$  is H or  $C_1$  to  $C_3$  alkyl;

$G_1$  is O,  $NR_7$ , or  $CR_7R_8$ ;

$G_2$  is CO, CS, or  $CR_7R_8$ ; provided that when  $G_1$  is O,  $G_2$  is  $CR_7R_8$ , and  $G_1$  and  $G_2$  cannot both be  $CR_7R_8$ ;

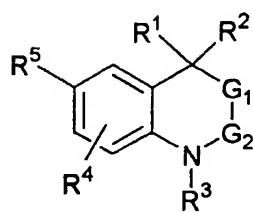


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$R_7$  and  $R_8$  are independently selected from H or an optionally substituted alkyl, aryl, or heterocyclic moiety;

or pharmaceutically acceptable salt thereof.

2. A compound of Claim 1 of Formula I:



I

wherein:

$R^1$  is H,  $C_1$  to  $C_6$  alkyl, substituted  $C_1$  to  $C_6$  alkyl,  $C_3$  to  $C_8$  cycloalkyl, substituted  $C_3$  to  $C_8$  cycloalkyl, aryl, substituted aryl, heterocyclic, substituted heterocyclic,  $COR^A$ , or  $NR^B COR^A$ ;

$R^2$  is H,  $C_1$  to  $C_6$  alkyl, substituted  $C_1$  to  $C_6$  alkyl,  $C_2$  to  $C_6$  alkenyl, substituted  $C_2$  to  $C_6$  alkenyl,  $C_3$  to  $C_8$  cycloalkyl, substituted  $C_3$  to  $C_8$  cycloalkyl, aryl, substituted aryl, heterocyclic, substituted heterocyclic,  $COR^A$ , or  $NR^B COR^A$ ;

or

$R^1$  and  $R^2$  are fused to form an optionally substituted 3 to 8 membered spirocyclic alkyl, alkenyl or heterocyclic ring containing one to three heteroatoms from the group including O, S and N, the spirocyclic alkyl, alkenyl or heterocyclic rings being optionally substituted by from 1 to 4 groups selected from fluorine,  $C_1$  to  $C_6$  alkyl,  $C_1$  to  $C_6$  alkoxy,  $C_1$  to  $C_6$  thioalkyl,  $-CF_3$ ,  $-OH$ ,  $-CN$ ,  $NH_2$ ,  $-NH(C_1$  to  $C_6$  alkyl), or  $-N(C_1$  to  $C_6$  alkyl) $_2$ ;

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$R^A$  is H,  $C_1$  to  $C_3$  alkyl, substituted  $C_1$  to  $C_3$  alkyl, aryl, substituted aryl,  $C_1$  to  $C_3$  alkoxy, substituted  $C_1$  to  $C_3$  alkoxy,  $C_1$  to  $C_3$  aminoalkyl, or substituted  $C_1$  to  $C_3$  aminoalkyl;

$R^B$  is H,  $C_1$  to  $C_3$  alkyl, or substituted  $C_1$  to  $C_3$  alkyl,

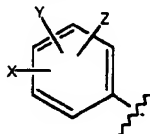
$R^3$  is H, OH,  $NH_2$ ,  $C_1$  to  $C_6$  alkyl, substituted  $C_1$  to  $C_6$  alkyl,  $C_3$  to  $C_6$  alkenyl, substituted  $C_1$  to  $C_6$  alkenyl, alkynyl, or substituted alkynyl,  $COR^C$ ;

$R^C$  is H,  $C_1$  to  $C_4$  alkyl, substituted  $C_1$  to  $C_4$  alkyl, aryl, substituted aryl,  $C_1$  to  $C_4$  alkoxy, substituted  $C_1$  to  $C_4$  alkoxy,  $C_1$  to  $C_4$  aminoalkyl, or substituted  $C_1$  to  $C_4$  aminoalkyl;

$R^4$  is H, halogen, CN,  $NO_2$ ,  $C_1$  to  $C_6$  alkyl, substituted  $C_1$  to  $C_6$  alkyl,  $C_1$  to  $C_6$  alkoxy, substituted  $C_1$  to  $C_6$  alkoxy, amino,  $C_1$  to  $C_6$  aminoalkyl, substituted  $C_1$  to  $C_6$  aminoalkyl,

$R^5$  is a) or b):

a)  $R^5$  is a benzene ring of the structure:



$X$  is selected from halogen, CN,  $C_1$  to  $C_3$  alkyl, substituted  $C_1$  to  $C_3$  alkyl,  $C_1$  to  $C_3$  alkoxy, substituted  $C_1$  to  $C_3$  alkoxy,  $C_1$  to  $C_3$  thioalkoxy, substituted  $C_1$  to  $C_3$  thioalkoxy, amino,  $C_1$  to  $C_3$  aminoalkyl, substituted  $C_1$  to  $C_3$  aminoalkyl,  $NO_2$ ,  $C_1$  to  $C_3$

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perfluoroalkyl, 5 membered heterocyclic ring containing 1 to 3 heteroatoms,  $\text{COR}^D$ ,  $\text{OCOR}^D$ , or  $\text{NR}^E\text{COR}^D$ ;

$\text{R}^D$  is H,  $\text{C}_1$  to  $\text{C}_3$  alkyl, substituted  $\text{C}_1$  to  $\text{C}_3$  alkyl, aryl, substituted aryl,  $\text{C}_1$  to  $\text{C}_3$  alkoxy, substituted  $\text{C}_1$  to  $\text{C}_3$  alkoxy,  $\text{C}_1$  to  $\text{C}_3$  aminoalkyl, or substituted  $\text{C}_1$  to  $\text{C}_3$  aminoalkyl;

$\text{R}^E$  is H,  $\text{C}_1$  to  $\text{C}_3$  alkyl, or substituted  $\text{C}_1$  to  $\text{C}_3$  alkyl;

Y and Z are independent substituents taken from the group including H, halogen, CN,  $\text{NO}_2$ ,  $\text{C}_1$  to  $\text{C}_3$  alkoxy,  $\text{C}_1$  to  $\text{C}_3$  alkyl, or  $\text{C}_1$  to  $\text{C}_3$  thioalkoxy;  
or

b)  $\text{R}^5$  is a five or six membered ring with 1, 2, or 3 heteroatoms selected from the group of O, S, SO,  $\text{SO}_2$  or  $\text{NR}^6$  and containing one or two independent substituents selected from H, halogen, CN,  $\text{NO}_2$ , amino, and  $\text{C}_1$  to  $\text{C}_3$  alkyl,  $\text{C}_1$  to  $\text{C}_3$  alkoxy;

$\text{R}^6$  is H, or  $\text{C}_1$  to  $\text{C}_3$  alkyl.

$\text{G}_1$  is O,  $\text{NR}_7$ , or  $\text{CR}_7\text{R}_8$

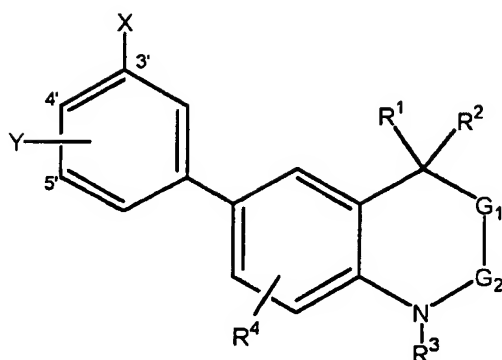
$\text{G}_2$  is CO, CS, or  $\text{CR}_7\text{R}_8$ ; provided that when  $\text{G}_1$  is O,  $\text{G}_2$  is  $\text{CR}_7\text{R}_8$ , and  $\text{G}_1$  and  $\text{G}_2$  cannot both be  $\text{CR}_7\text{R}_8$ ;

$\text{R}_7$  and  $\text{R}_8$  are independently selected from H, alkyl, substituted alkyl, aryl, substituted aryl, heterocyclic, or substituted heterocyclic;

or a pharmaceutically acceptable salt thereof.

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3. A compound of Claim 1 having the formula:



wherein:

$R^1 = R^2$  and are selected from  $C_1$  to  $C_3$  alkyl, substituted  $C_1$  to  $C_3$  alkyl, or spirocyclic alkyl constructed by fusing  $R^1$  and  $R^2$  to form a 3 to 6 membered spirocyclic ring, the spirocyclic ring being optionally substituted by from 1 to 4 groups selected from fluorine,  $C_1$  to  $C_6$  alkyl,  $C_1$  to  $C_6$  alkoxy,  $C_1$  to  $C_6$  thioalkyl,  $-CF_3$ ,  $-OH$ ,  $-CN$ ,  $NH_2$ ,  $-NH(C_1 \text{ to } C_6 \text{ alkyl})$ , or  $-N(C_1 \text{ to } C_6 \text{ alkyl})_2$ ;

$R^3$  is H, OH,  $NH_2$ ,  $C_1$  to  $C_6$  alkyl, substituted  $C_1$  to  $C_6$  alkyl,  $-COH$ ,  $-CO(C_1 \text{ to } C_4 \text{ alkyl})$  or  $-CO(C_1 \text{ to } C_4 \text{ alkoxy})$ ;

$R^4$  is H, halogen,  $NO_2$ ,  $C_1$  to  $C_3$  alkyl, substituted  $C_1$  to  $C_3$  alkyl;

X is taken from the group including halogen, CN,  $C_1$  to  $C_3$  alkoxy,  $C_1$  to  $C_3$  alkyl,  $NO_2$ ,  $C_1$  to  $C_3$  perfluoroalkyl, 5 membered heterocyclic ring containing 1 to 3 heteroatoms,  $C_1$  to  $C_3$  thioalkoxy,

Y is a substituent on the 4' or 5' position from the group including H, halogen, CN,  $NO_2$ ,  $C_1$  to  $C_3$  alkoxy,  $C_1$  to  $C_4$  alkyl,  $C_1$  to  $C_3$  thioalkoxy

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$G_1$  is O,  $NR_7$ , or  $CR_7R_8$

$G_2$  is CO, CS, or  $CR_7R_8$ ; provided that when  $G_1$  is O,  $G_2$  is  $CR_7R_8$ , and  $G_1$  and  $G_2$  cannot both be  $CR_7R_8$ ;

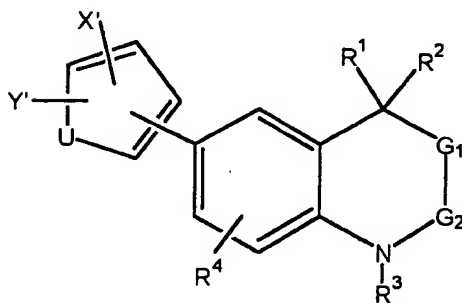
wherein  $R_7$  and  $R_8$  are independent substituents selected from H, alkyl, substituted alkyl, aryl, substituted aryl, heterocyclic, or substituted heterocyclic;

or pharmaceutically acceptable salt thereof

4. A compound of Claim 3 wherein X is CN and Y is 5'-fluoro; or a pharmaceutically acceptable salt thereof.

5. A compound of Claim 3 wherein U is S,  $X'$  is CN and  $Y'$  is H or  $-CH_3$ ; or a pharmaceutically acceptable salt thereof.

6. A compound of Claim 1 having the formula:



wherein:

$R^1 = R^2$  and are selected from  $C_1$  to  $C_3$  alkyl, substituted  $C_1$  to  $C_3$  alkyl, or spirocyclic alkyl constructed by fusing  $R^1$  and  $R^2$  to form a 3 to 6 membered

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spirocyclic ring, the spirocyclic ring being optionally substituted by from 1 to 4 groups selected from fluorine, C<sub>1</sub> to C<sub>6</sub> alkyl, C<sub>1</sub> to C<sub>6</sub> alkoxy, C<sub>1</sub> to C<sub>6</sub> thioalkyl, -CF<sub>3</sub>, -OH, -CN, NH<sub>2</sub>, -NH(C<sub>1</sub> to C<sub>6</sub> alkyl), or -N(C<sub>1</sub> to C<sub>6</sub> alkyl)<sub>2</sub>;

R<sup>3</sup> is H, OH, NH<sub>2</sub>, C<sub>1</sub> to C<sub>6</sub> alkyl, substituted C<sub>1</sub> to C<sub>6</sub> alkyl, -COH, -CO(C<sub>1</sub> to C<sub>4</sub> alkyl) or -CO(C<sub>1</sub> to C<sub>4</sub> alkoxy);

R<sup>4</sup> is H, halogen, NO<sub>2</sub>, C<sub>1</sub> to C<sub>3</sub> alkyl, substituted C<sub>1</sub> to C<sub>3</sub> alkyl;

U is O, S, or NR<sup>6</sup>,

R<sup>6</sup> is H, or C<sub>1</sub> to C<sub>3</sub> alkyl, C<sub>1</sub> to C<sub>4</sub> CO<sub>2</sub>alkyl,

X' is from the group including halogen, CN, NO<sub>2</sub>, C<sub>1</sub> to C<sub>3</sub> alkyl and C<sub>1</sub> to C<sub>3</sub> alkoxy;

Y' is from the group including H and C<sub>1</sub> to C<sub>4</sub> alkyl;

G<sub>1</sub> is O, NR<sub>7</sub>, or CR<sub>7</sub>R<sub>8</sub>;

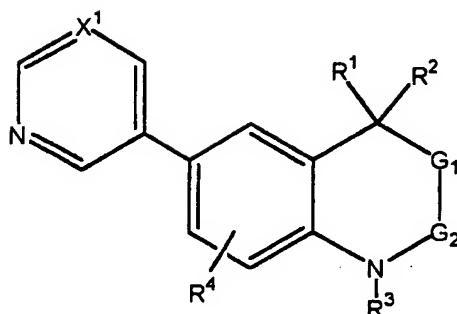
G<sub>2</sub> is CO, CS, or CR<sub>7</sub>R<sub>8</sub>; provided that when G<sub>1</sub> is O, G<sub>2</sub> is CR<sub>7</sub>R<sub>8</sub>, and G<sub>1</sub> and G<sub>2</sub> cannot both be CR<sub>7</sub>R<sub>8</sub>;

wherein R<sub>7</sub> and R<sub>8</sub> are independent substituents selected from H, alkyl, substituted alkyl, aryl, substituted aryl, heterocyclic, or substituted heterocyclic;

or pharmaceutically acceptable salt thereof

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7. A compound of Claim 1 having the formula:



wherein:

$R^1 = R^2$  and are selected from  $C_1$  to  $C_3$  alkyl, substituted  $C_1$  to  $C_3$  alkyl, or spirocyclic alkyl constructed by fusing  $R^1$  and  $R^2$  to form a 3 to 6 membered spirocyclic ring, the spirocyclic rings of being optionally substituted by from 1 to 4 groups selected from fluorine,  $C_1$  to  $C_6$  alkyl,  $C_1$  to  $C_6$  alkoxy,  $C_1$  to  $C_6$  thioalkyl,  $-CF_3$ ,  $-OH$ ,  $-CN$ ,  $NH_2$ ,  $-NH(C_1 \text{ to } C_6 \text{ alkyl})$ , or  $-N(C_1 \text{ to } C_6 \text{ alkyl})_2$ ;

$R^3$  is H, OH,  $NH_2$ ,  $C_1$  to  $C_6$  alkyl, substituted  $C_1$  to  $C_6$  alkyl,  $-COH$ ,  $-CO(C_1 \text{ to } C_4 \text{ alkyl})$  or  $-CO(C_1 \text{ to } C_4 \text{ alkoxy})$ ;

$R^4$  is H, halogen,  $NO_2$ ,  $C_1$  to  $C_3$  alkyl, substituted  $C_1$  to  $C_3$  alkyl;

$X^1$  is N or  $CX^2$ ,

$X^2$  is halogen, CN, alkoxy, or  $NO_2$ .

$G_1$  is O,  $NR_7$ , or  $CR_7R_8$

$G_2$  is CO, CS, or  $CR_7R_8$ ; provided that when  $G_1$  is O,  $G_2$  is  $CR_7R_8$ , and  $G_1$  and  $G_2$  cannot both be  $CR_7R_8$ ;

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wherein R<sub>7</sub> and R<sub>8</sub> are independent substituents selected from H, alkyl, substituted alkyl, aryl, substituted aryl, heterocyclic, or substituted heterocyclic;

or pharmaceutically acceptable salt thereof

8. A compound of Claim 1 which is 6-(3-Chloro-phenyl)-2,4,4-trimethyl-2-trifluoromethyl-1,4-dihydro-2H-benzo[d][1,3]oxazine or a pharmaceutically acceptable salt thereof.

9. A compound of Claim 1 which is 6-(3-Chloro-phenyl)-2,2,4,4-tetramethyl-1,4-dihydro-2H-benzo[d][1,3]oxazine or a pharmaceutically acceptable salt thereof.

10. A compound of Claim 1 which is 6-(3-Nitro-phenyl)-2,2,4,4-trimethyl-1,4-dihydro-2H-benzo[d][1,3]oxazine or a pharmaceutically acceptable salt thereof.

11. A compound of Claim 1 which is 4-Amino-3'-chloro-biphenyl-3-carbonitrile or a pharmaceutically acceptable salt thereof.

12. A compound of Claim 1 which is 6-(3-Chlorophenyl)-4-cyclopropyl-1H-quinazolin-2-one or a pharmaceutically acceptable salt thereof.

13. A compound of Claim 1 which is 6-(3-Chlorophenyl)-4-cyclopropyl-1-(4-methoxybenzyl)-1H-quinazolin-2-one or a pharmaceutically acceptable salt thereof.

14. A compound of Claim 1 which is 6-(3-Chlorophenyl)-4-cyclopropyl-4-methyl-3,4-dihydro-1H-quinazolin-2-one or a pharmaceutically acceptable salt thereof.



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15. A compound of Claim 1 which is 6-(3-Chlorophenyl)-4-cyclopropyl-3,4-dimethyl-3,4-dihydro-1H-quinazolin-2-one or a pharmaceutically acceptable salt thereof.

16. A compound of Claim 1 which is 6-(3-Chloro-phenyl)-4,4-dimethyl-3,4-dihydro-1H-quinolin-2-one or a pharmaceutically acceptable salt thereof.

17. A pharmaceutical composition comprising a compound of Claim 1, or a pharmaceutically acceptable salt thereof, and a pharmaceutically acceptable carrier or excipient.

18. A method of inducing contraception in a mammal, the method comprising administering to a mammal in need thereof a pharmaceutically effective amount of a compound of Claim 1 or a pharmaceutically effective amount thereof.

19. A method of treating or preventing benign or malignant neoplastic disease in a mammal, the method comprising administering to a mammal in need thereof a pharmaceutically effective amount of a compound of Claim 1 wherein  $G_2$  is  $C=O$  or  $CR_7R_8$ , or a pharmaceutically acceptable salt thereof.

20. The method of Claim 19 wherein the benign or malignant neoplastic disease is selected from uterine myometrial fibroids, endometriosis, benign prostatic hypertrophy; carcinomas and adenocarcinomas of the endometrium, ovary, breast, colon, prostate, pituitary, meningioma or other hormone-dependent tumors.

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21. A method of treatment or prevention in a mammal of carcinomas or adenocarcinomas of the endometrium, ovary, breast, colon, or prostate, the method comprising administering to a mammal in need thereof a pharmaceutically effective amount of a compound of Claim 1 wherein  $G_2$  is  $C=S$ , or a pharmaceutically acceptable salt thereof.

# INTERNATIONAL SEARCH REPORT

Int. l. Application No.

PCT/US 00/11835

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C07D215/22 C07D413/04 C07D239/80 C07D265/14 C07D239/78  
A61K31/517 A61P5/00

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C07D A61K A61P

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

CHEM ABS Data

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 96 19458 A (LIGAND PHARMACEUTICALS) 27 June 1996 (1996-06-27) cited in the application claims; examples 43,56,351-5	1-4,17

☐ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

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Date of the actual completion of the international search

29 August 2000

Date of mailing of the international search report

07/09/2000

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Authorized officer

Francois, J

# INTERNATIONAL SEARCH REPORT

Information on patent family members

In. ational Application No

PCT/US 00/11835

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